

Chemistry SUB. CODE- BSCC2402

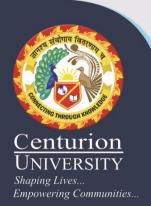
For B.Sc. 4th Semester

(Heterocycles) Module-IV

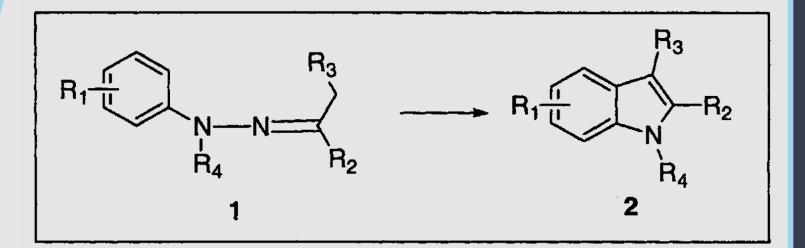
Indole: Synthesis

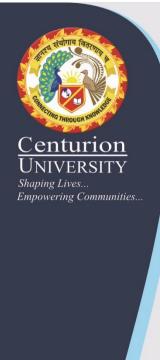
By

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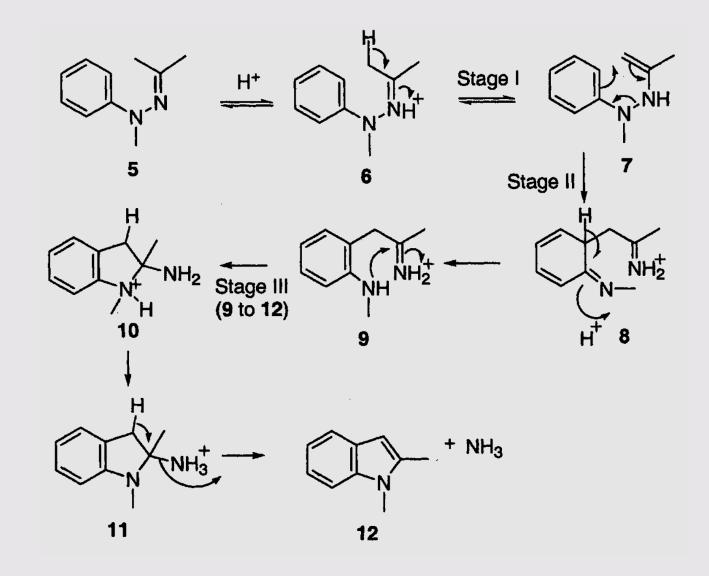


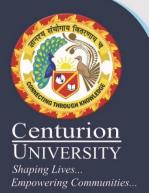
The Fischer indole synthesis can be regarded as the cyclization of an arylhydrazone **1** of an aldehyde or ketone by treatment with acid catalyst or effected thermally to form the indole nucleus.





Mechanism





Mechanism

Step 1: Under acidic conditions, protonation of the imine nitrogen followed by tautomerization to form an ene-hydrazine intermediate (7).

Step 2: After the tautomerization, **a [3,3]** -sigmatropic rearrangement occurs, which provides intermediate **8**.

Step 3: Rearomatization then occurs *via* a proton shift to form the imine **9** which cyclizes to form the 5-membered ring **10.** Finally, loss of ammonia from **11** generates the indole nucleus in **12.**



Commonly used catalysts for the Fischer indole synthesis

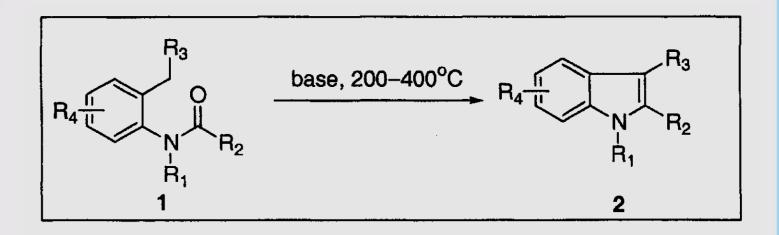
Glacial acetic acid Boron trifluoride in acetic acid Copper (I) chloride Formic acid Conc. hvdrochloric acid Hydrogen chloride in ethanol Hydrogen chloride in acetic acid Polyphosphoric acid Conc. sulfuric acid in ethanol Conc. sulfuric acid in acetic acid Zinc chloride p-Toluenesulfonic acid



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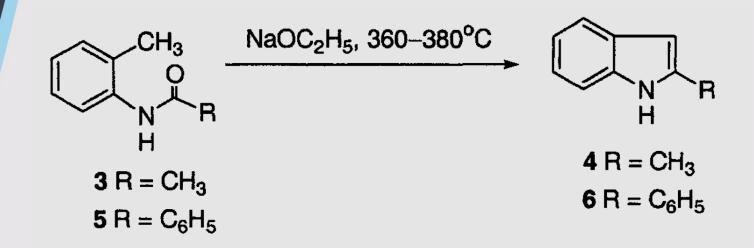
Madelung Indole Synthesis

The intramolecular cyclization of **N-acylated-o-alkylanilines** in the presence of a strong base at elevated temperatures is known as the Madelung indole synthesis





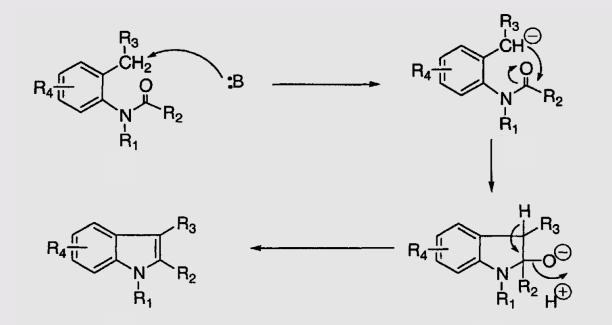
First reported Madelung indole synthesis

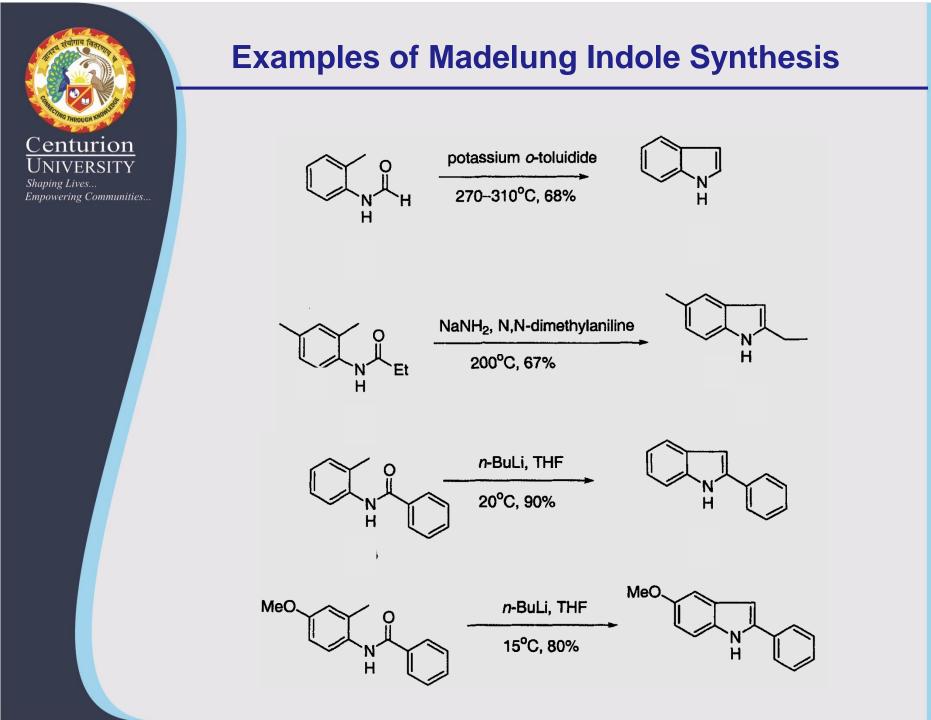




Mechanism

The mechanism of the Madelung indole synthesis has not been fully established. Presumably it involves an intramolecular Claisen type condensation.







References

- 1. K.D Sharma and Y. R. Sharma, Kalyani Publishers, Unit-III page-278;
- 2. University Chemistry, Vol-IV, Dr. U. N. Dash, Dr. KK. Ojha, Himalaya Publishing house, Unit IV, page-378