

ASAC 1101 : FUNDAMENTALS OF SOIL SCIENCE (2+1)

Level : B.Sc (Ag), I semester

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Topic Soil Organic Matter

Organic Matter

- Soil organic matter is the fraction of the soil that consists of plant or animal residues in various stages of breakdown (decomposition).
- In addition to these material, Soil organic matter consist of living and dead microbial cells, microbial synthesized compounds and a number of derivatives produced as a result of microbial activity.

In general, cultivated soil in temperate regions has high OM levels (5-10%), similarly in tropics have only less than 1%

Sources of OM

- SOM mainly originates from the plant tissue.
- Leaves and roots of trees, shrubs, grasses and other plants etc.
- These plant constituents form primary material both for food of soil organisms and for the production of soil organic matter.
- Animals are generally considered as secondary sources of SOM.

Composition of green plant material

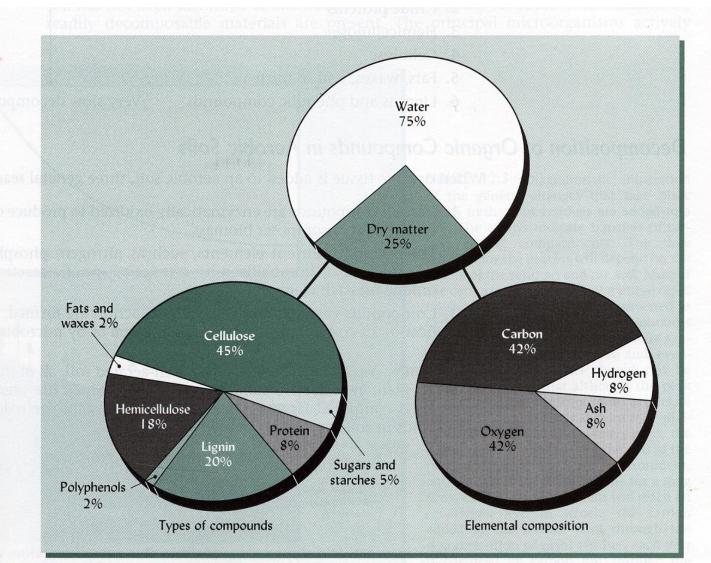


FIGURE 12.2 Typical composition of representative green-plant materials. The major types of organic compounds are indicated at left and the elemental composition at right. The *ash* is considered to include all the constituent elements other than carbon, oxygen, and hydrogen (nitrogen, sulfur, calcium, etc.).

Role of organic matter

A. Physical Benefits

- Enhances aggregate stability, improving water infiltration and soil aeration, reducing runoff.
- Improves water holding capacity.
- Creates a granular condition of soil which maintains favorable condition of aeration and permeability.
- Reduces surface crusting, facilitating seedbed preparation.
- Surface mulching with coarse organic matter lowers wind erosion and lowers soil temperatures in the summer and keeps the soil warmer in winter.

B. Chemical Benefits

- Increases the soil's CEC or its ability to hold and supply essential nutrients.
- Improves the ability of a soil to resist pH change; this is also known as buffering capacity.

• Humus (a highly decomposed organic matter) provides a storehouse for the exchangeable and available cations.

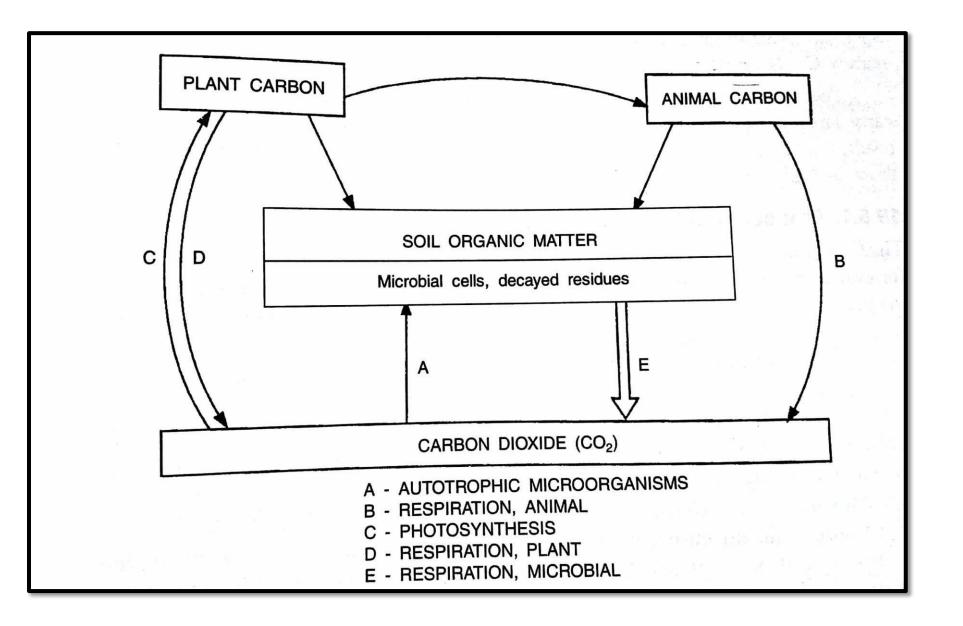
• Organic acids released from decomposing organic matter help to reduce alkalinity in soils.

C. Biological Benefits

- Organic matter serves as a source of energy for the microbes.
- Fresh Organic matter supplies food for earthworms, ants and rodents.
- Enhances soil microbial biodiversity and activity which can help in the suppression of diseases and pests.

CARBON CYCLE

- The series of processes by which carbon compounds are inter converted into the environment, involving the incorporation of carbon dioxide into living tissue by photosynthesis and its return to the atmosphere through respiration, the decay of dead organisms, and the burning of fossil fuels.
- The carbon cycle revolves about CO₂ and its fixation and regeneration.
 The carbon cycle was discovered by Joseph Priestley and Antoine Lavoisier, and popularized by Humphry Davy.



Steps of the Carbon Cycle:

- **1.** CO₂ used by Plants for Photosynthesis
- **2.** Consumption by Animals
- The carbon present in the food made by green plants reaches animals through the food chain. Carnivorous animals receive this carbon when they eat other animals.

3. Respiration by Plants, Animals and Microbes

• Carbon dioxide is regularly being returned to the atmosphere by the process of respiration by plants, animals and microbes.

4. Decay and Decomposition

• When living organisms die, their bodies decay and decompose.

Decomposition of soil organic matter:

- The term decomposition means "to break down". It usually refers to the rupture or breakdown of the complex organic matter into a simpler inorganic matter.
- The organic matter is also classified on the basis of their rate of decomposition
 - **1. Rapidly decomposed** : Sugars, starches, proteins *etc*.
 - 2. Less rapidly decomposed : Hemicelluloses, celluloses etc.
 - 3. Very slowly decomposed : Fats, waxes, resins, lignins etc.

The general reactions taking place during

decomposition are

- \circ Enzymatic oxidation with the release of Co_2 , water, energy and heat
- Essential elements are released (N, P, S etc) and immobilized by a series of reactions.
- Formation of compounds which are resistant to microbial action.

Under aerobic conditions the products formed are
CO₂, NH₄, NO₃, H₂PO₄, SO₄, H₂O and essential
plant nutrients like Ca, Mg, Fe, Cu, Zn etc.

Under anaerobic conditions

• CH₄, organic acids like lactic, butyric, NH₄, various amine residues (R-NH₂) H₂S and ethylene (CH₂=CH₂). A. Decomposition of soluble substances
Aerobic conditions
Sugar + Oxygen → CO₂ + H₂O
Under partially oxidized conditions,
Sugar + Oxygen → Aliphatic acids (Acetic, formic *etc.*) or
acids (Citric, lactic) etc. or Alcohols (ethyl alcohol) etc.

B. Decomposition of Insoluble Substances

i. Breakdown of Protein:

Proteins are first hydrolyzed to a number of intermediate products.

Aminization: The process of conversion of proteins to amino acids.
 Ammonification: The transformation of organic nitrogenous compounds (amino acids and amides) into ammonia is called ammonification.

Nitrification: The process of conversion of ammonia to nitrites (NO_2^-) and then to nitrate (NO_3^-) is known as nitrification.

• It is an aerobic process by autotrophic bacteria.

• Nitrification of ammonia to nitrate is a two step process:

Nitrosomonas Nitrobacter $NH_4^+ \longrightarrow NO_2^- \longrightarrow NO_3^-$ (Nitrite) (Nitrate)

- (Oxidation of ammonium in to nitrites and nitrites to nitrates).
- **Denitrification:** The process, which involves conversion of soil nitrate into gaseous nitrogen or nitrous oxide under anaerobic condition, is called Denitrification.

 $NO_3 \longrightarrow NO_2 \longrightarrow NO \longrightarrow N_2O \longrightarrow N_2$

ii) Breakdown of cellulose:

The decomposition of the most abundant carbohydrates.

cellulase cellobiase Cellulose \longrightarrow Cellobiose \longrightarrow Glucose \longrightarrow Organic acids \longrightarrow CO₂ + H₂O

iii)Breakdown of Hemicellulose:

- Decompose faster than cellulose and are first hydrolyzed to their components sugars.
- Sugars are attacked by microbes and are converted to organic acids, alcohols, carbon dioxide and water.

Breakdown of Starch:

• It is chemically a glucose polymer and is first hydrolyzed to maltose by the action of amylases. Maltose is next converted to glucose by maltase.

amylase maltase Starch → maltose → glucose

Decomposition of ether soluble substances:

Fats \longrightarrow glycerol + fatty acids Glycerol \longrightarrow CO₂ + water

Decomposition of lignin:

• Lignin decomposes slowly, much slower than cellulose. Complete oxidation gives rise to CO₂ and H₂O.

Factors affecting OM decomposition

- **1.Temperature:** The decomposition of organic residues is accelerated in warm temperature than that of cold climate. Decomposition of SOM increases with rise in T from 0-35 °C
- **2.Soil moisture:** Extremes of both arid and anaerobic conditions reduce plant growth and microbial decomposition.
- Near or slightly wetter than field capacity (60-80% of maximum water holding capacity) moisture conditions are most favorable for both processes.
- **3.Nutrients:** Lack of nutrients particularly N slows decomposition.
- **4.Soil PH:** Most of the microbes grow best at pH 6 8, but are severely inhibited below pH 4.5 and above pH 8.5.
- **5.Soil Texture:** Soils higher in clays tend to retain larger amounts of humus.

- **6.Type of vegetation:** OM levels of cultivated area are lower compared to forest and grassland, legumes are more easily decomposed than grasses.
- **7.Other Factors:** Toxic levels of elements (Al, Mn, B, Se, Cl), excessive soluble salts, shade and organic phytotoxins in plant materials.

Fractions of OM

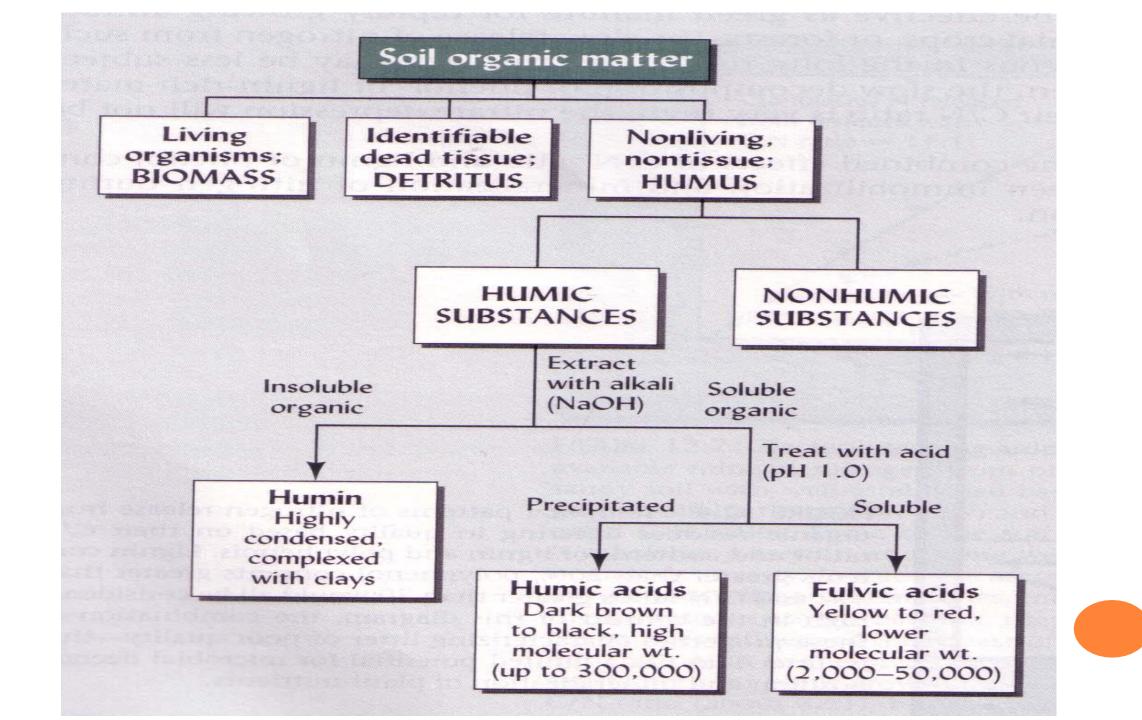
1.Humic group

2.Non humic group

Humic group

- This group makes up about **60-80%** of the soil organic matter.
- They are most complex. They are most resistant to microbial attack.
- The humic substances are classified based on resistance to degradation and solubility in acids and alkalis into

Humic acid Fulvic acid Humin



Fulvic acid	lowest in molecular weight, lightest in colour, soluble in both acid and alkali and most susceptible to microbial attack	
Humic acid	Medium in molecular weight and colour, soluble in alkali but insoluble in acid and intermediate in resistance to degradation	
Humin	Highest in molecular weight, darkest in colour, insoluble in both acid and alkali and most resistant to microbial attack	

Non humic group

- This group makes upto 20-30% of the organic matter in soil.
- These are less complex and less resistant to microbial attack as compared to humic substances.
- These include proteins, carbohydrates, lignins, fats, waxes, resins, tannins and some compounds of low molecular weight.

Humus

• Humus is a complex and rather resistant mixture of brown or dark brown amorphous and colloidal organic substance that results from microbial decomposition and has chemical and physical properties of great significance to soils and plants.

Properties of Humus

- The tiny colloidal particles are composed of C,H, and O
- The colloidal particles are negatively charged (-OH, -COOH or phenolic groups).
- Very high surface area.
- Higher CEC (150 300 cmol/kg).
- 4 -5 times higher WHC than that of silicate clays.
- Humus has a very favorable effect on aggregate formation and stability.
- Impart black colour to soils.

Theories on humus formation

Lignin theory: Proposed by Waksman (1936).

 According to this theory humic substances are formed due to the incomplete degradation of lignin.

Kononovas theory:

 According to this theory humic substances are formed by cellulose decomposing mycobacteria earlier to lignin decomposition.

Polyphenol theory: (Flaig and Sochtig (1964)

- As per this theory the humic substances are formed by the condensation of phenolic materials.
- The polyphenols of lignin are oxidized to quinones.
- These quinones are condensed with low molecular weight microbial products to form humic molecules.



C: N RATIO

• The ratio between the nitrogen content in the microbes and in the organic residues and to the carbon content is called as C:N ratio.

Importance of C:N ratio

- C:N ratio mainly controls decomposition rate in soil.
- The wide C:N ratio leads to slow decomposition rate, nutrient **immobilization** may occur, carbon and energy wastage in large quantities.
- In Narrow C:N ratio, Plant residues decompose quickly and release nitrates readily.
- When the residue with high C/N ratio is added to soils, there will be intense competition among the microorganism for available N.
- The C/N ratio in residues helps determine their rate of decay and the rate at which N is made available to plants.

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	C:N ratio
Indian Soil	8.5-12:1
Cultivated soil	8-15:1
Humus	10:1
Microorganisms	4-9:1
FYM	100:1
Legumes	10-30:1
Sawdust	600:1

THANK YOU