Principles of protein structures

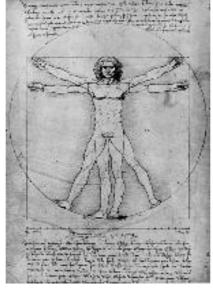
Ceci n'est pas une pomme

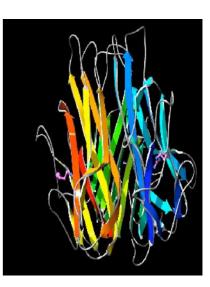














Amino Acids and Polypeptides (1)

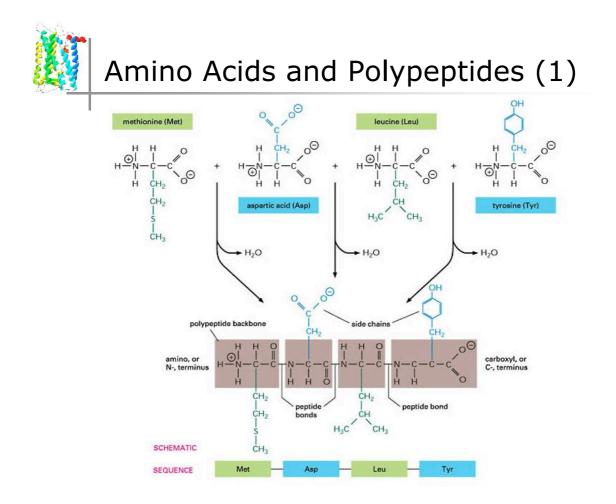
Physicochemical Properties of Amino Acid Side Chains (2)

Protein Folding and Physicochemical Interactions (3)

Principles of Protein Structure (4)

Structure Comparison (5)

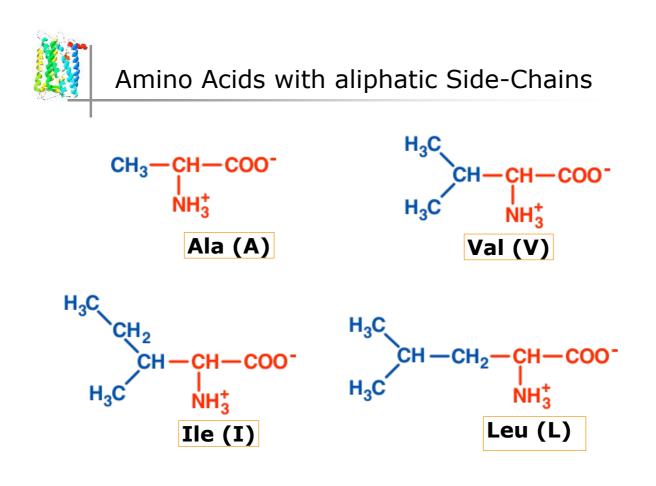
Protein Structure Databases (6)



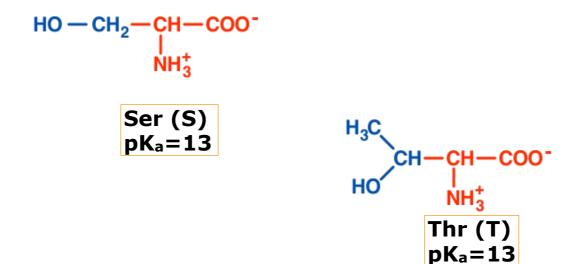


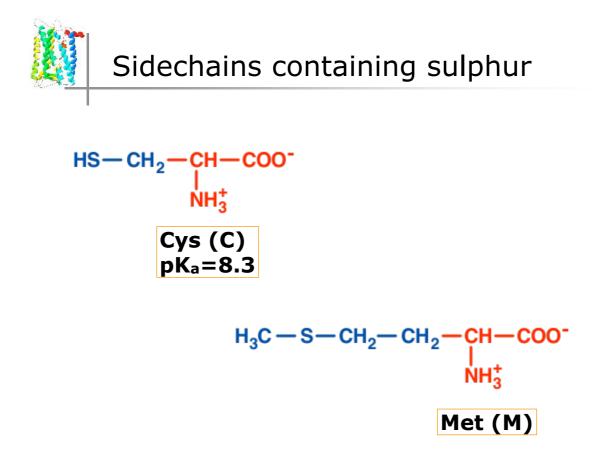
Three and one letter code:

| Alanine | Ala | А |
|-----------------------------|-----|---|
| Arginine | Arg | R |
| Asparagine | Asn | N |
| Aspartic acid | Asp | D |
| Asparagine or aspartic acid | Asx | В |
| Cysteine | Суз | С |
| Glutamine | Gin | Q |
| Glutamic acid | Glu | E |
| Glutamine or glutamic acid | GIx | Z |
| Glycine | Gly | G |
| Histidine | His | н |
| soleucine | lle | 1 |
| Loudine | Leu | L |
| Lysine | Lys | к |
| Methionine | Met | М |
| Phenylalanine | Phe | F |
| Proline | Pro | Р |
| Serine | Ser | S |
| Threonina | Thr | т |
| Tryptophan | Тгр | w |
| Tyrosine | Tyr | Y |
| Valine | Val | v |

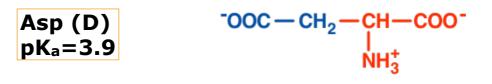


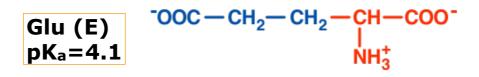


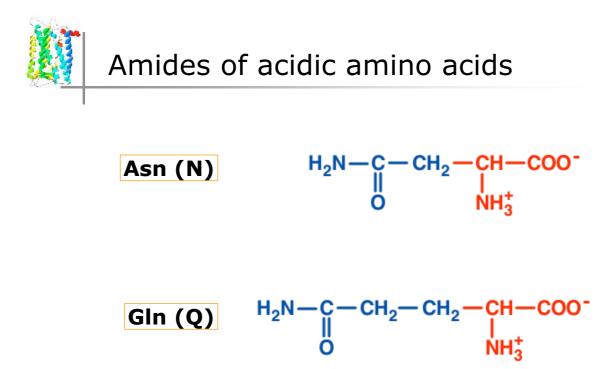






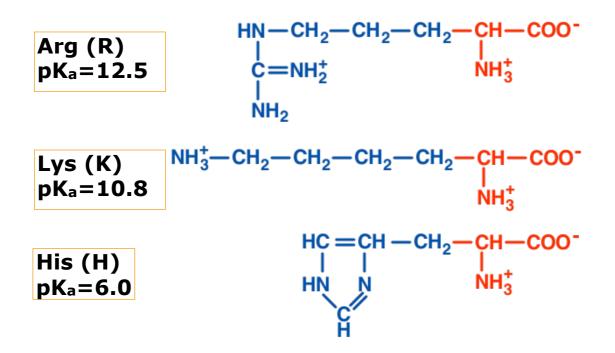


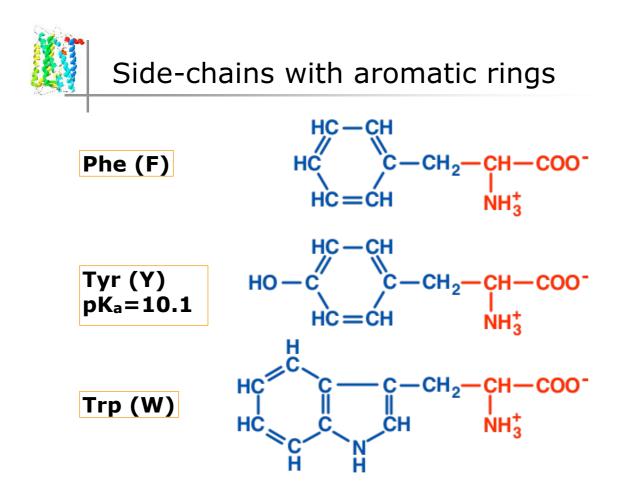




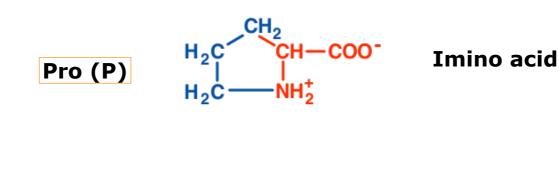


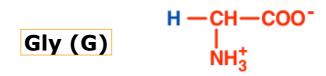
Basic Amino Acids

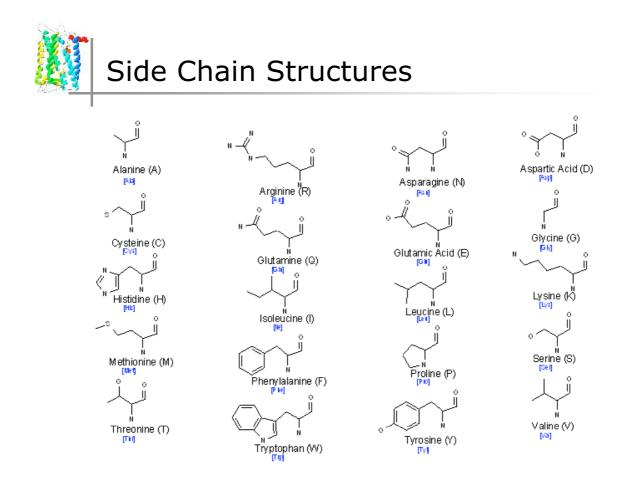














Physicochemical Properties of Amino Acid Side Chains (2)

Neutral Hydrophobic

Alanine Valine Leucine Isoleucine Proline Tryptophane Phenylalanine Methionine

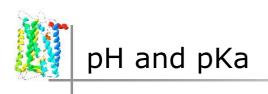
Neutral Polar

Glycine Serine Threonine Tyrosine Cysteine Asparagine Glutamine

Basic

Lysin Arginine (Histidine)

<u>Acidic</u> Aspartic Acid Glutamic Acid



D pH

 $pH = -\log [H^+]$

Water ion product

 $Kw = [H^+] [OH^-] = 10^{-14}$ $\log[H^+] + \log[OH^-] = \log(10^{-14})$ pH + pOH = 14



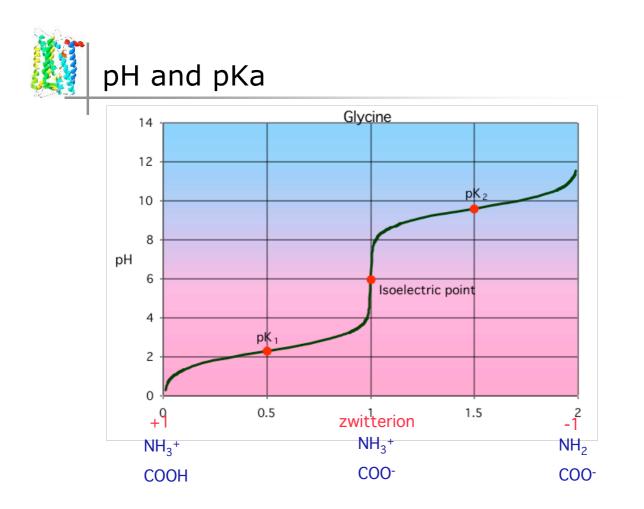
Dissociation of weak acids

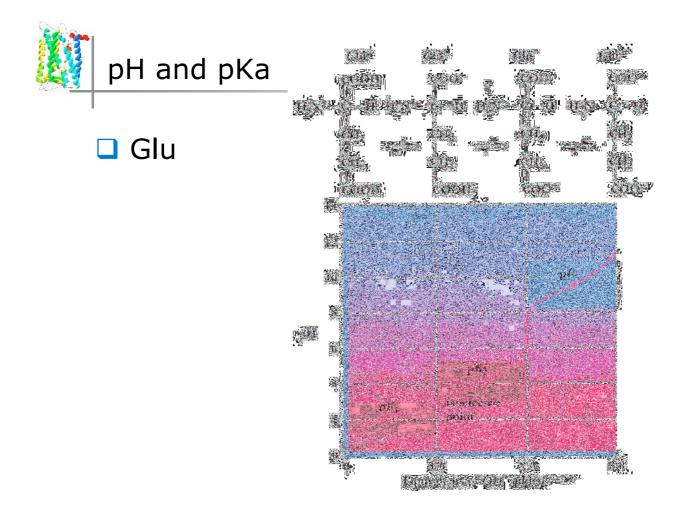
 $HA \iff H^{+} + A^{-}$ $K_{a} = \frac{[H^{+}][A^{-}]}{[HA]} \qquad [H^{+}] = K_{a} \frac{[HA]}{[A^{-}]}$ $\log [H^{+}] = \log K_{a} + \log \frac{[HA]}{[A^{-}]}$ $-\log [H^{+}] = -\log K_{a} + \log \frac{[A^{-}]}{[HA]}$

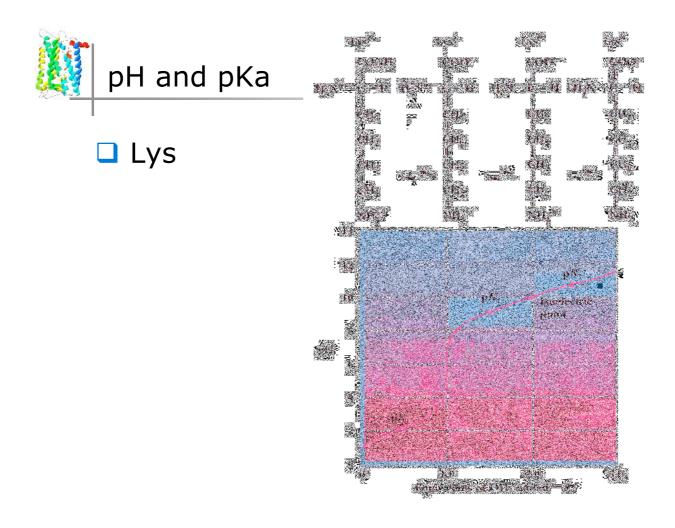


Henderson - Hasselbach Equation

$$pH = pKa + \log \frac{[A^-]}{[HA]}$$



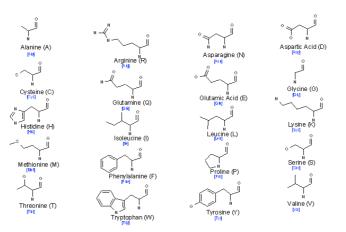


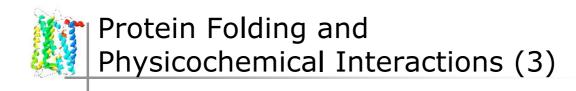




Enzymatic reactions often require proton transfer.

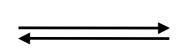
Q: Which amino-acid(s) are able to change their protonation state under physiological conditions?



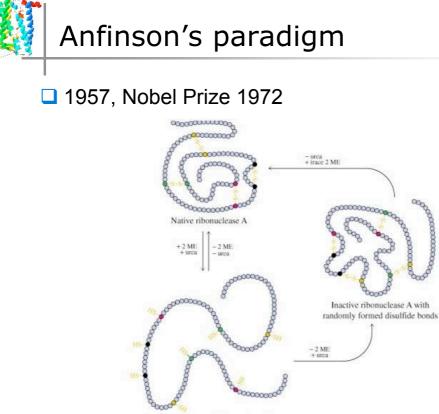


Why do proteins fold?

MNIFEMLRID EGLRLKIYKD TEGYYTIGIG HLLTKSPSLN AAKSELDKAI GRNCNGVITK DEAEKLFNQD VDAAVRGILR NAKLKPVYDS LDAVRRCALI NMVFQMGETG VAGFTNSLRM LQQKRMDEAA VNLAKSRWYN QTPNRAKRVI TTFRTGTWDA YKNL







Reversibly denatured ribonuclease A; disulfide bonds have been reduced



MNIFEMLRID EGLRLKIYKD TEGYYTIGIG HLITKSPSLN AAKSELDKAI GRNCNGVITK DEAEKLFNQD VDAAVRGILR NAKLKPVYDS LDAVRRCALI NMVFQMGETG VAGFTNSLRM LQQKRWDEAA VNLAKSRWYN QTPNRAKRVI TTFRTGTWDA YKNL



All the necessary information for the 3-dimensional structure of an enzyme is contained in the primary structure or sequence of the amino acids.



Levinthal's Paradox (1968)

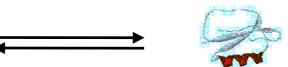
If a chain of a hundred amino acids is considered and it assumed each amino acid can exist in one of three conformations, extended, helical or loop, then there are 3¹⁰⁰ possible ways to arrange this chain.

This is roughly 10^{48} conformations. Bond rotation can be estimated to occur at a rate of roughly 10^{14} s⁻¹. This means that search for the right conformation through random searching alone would take the order of 10^{34} s or 10^{26} years, several orders of magnitudes greater than the age of the universe!

□ [J. Chim. Phys., **1968**, 85, 44]



| - | | | | |
|---|------------|------------|------------|---|
| | MNIFEMLRID | EGLRLKIYKD | TEGYYTIGIG | |
| | HLLTKSPSLN | AAKSELDKAI | GRNCNGVITK | |
| | DEAEKLFNQD | VDAAVRGILR | NAKLKPVYDS | _ |
| | LDAVRRCALI | NMVFQMGETG | VAGFTNSLRM | |
| | LQQKRWDEAA | VNLAKSRWYN | QTPNRAKRVI | |
| | TTFRTGTWDA | YKNL | | |
| | | | | |

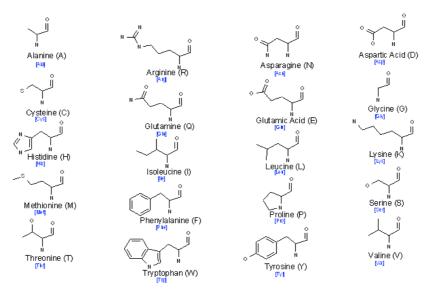


- Many proteins fold spontaneously to their native structure
- □ Protein folding is relatively fast (nsec sec)
- □ Chaperones speed up folding, but do not alter the structure

The protein sequence contains all information needed to create a correctly folded protein.



- N - $C_{\alpha}(HR_1)$ - CO - N - $C_{\alpha}(HR_2)$ - CO - N - $C_{\alpha}(HR_1)$ - CO -





Neutral Hydrophobic

Alanine Valine Leucine Isoleucine Proline Tryptophane Phenylalanine Methionine

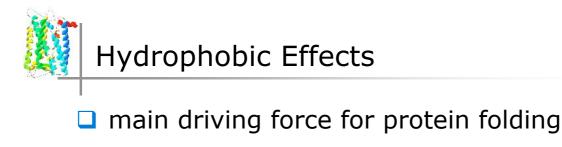
Neutral Polar

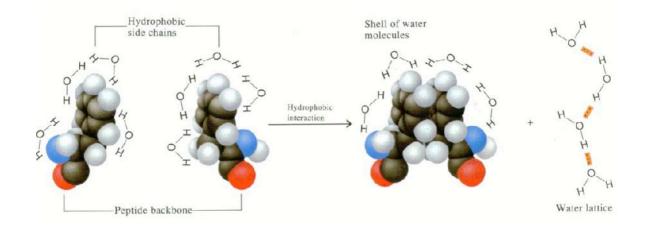
Glycine Serine Threonine Tyrosine Cysteine Asparagine Glutamine

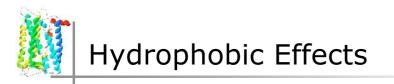
Basic

Lysin Arginine (Histidine)

<u>Acidic</u> Aspartic Acid Glutamic Acid

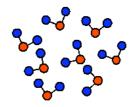


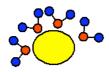




main driving force for protein folding

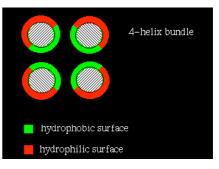
Water molecules in bulk water are mobile and can form H-bonds in all directions.

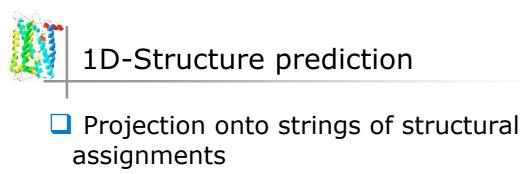




Hydrophobic surfaces don't form H-bonds. The surrounding water molecules have to orient and become more ordered.

The entropy loss can be minimized by gathering the hydrophobic surfaces together in the core of a protein and separating them from the solvent.



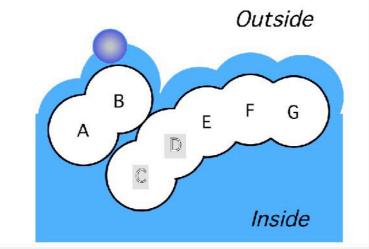


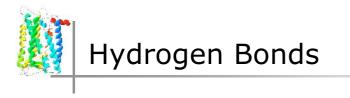
E.g. "Solvent Accessibility"

A B C D E F G... | | | | | | | e e b b e e e...

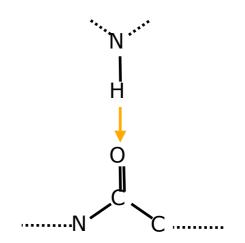


- Van der Waals Radius
- Molecular Surface
- Solvent Accessible Surface



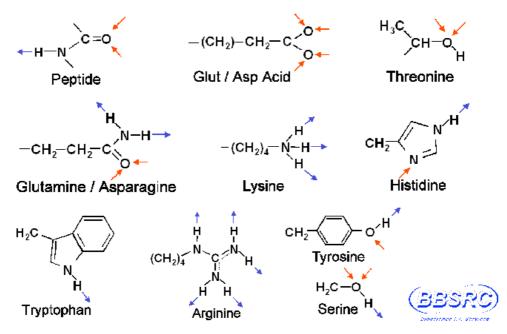


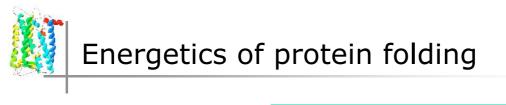
- H-atoms bound to electronegative atoms (e.g. N, O) are polarized and can form H-bonds
- □ H-bonding partners include:
 - main chain atoms
 - side chain atoms
 - water molecules
 - > ligands, etc...





Hydrogen bonding amino acids





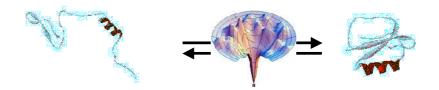
 $\Delta G = \Delta H - T \Delta S$

Energetics of protein folding

H-bonds hydrophobic effects salt bridges SS - bonds loss of solvation entropy change dispersion / VdW contacts conformational energy

Difference of two very large energetic terms
Low overall stabilization energy





- □ Change of energy state from unfolded to folded
- Folded state must have overall lower energy
- Let's assume the folded state is the lowest possible state for this polypeptide



 How many different proteins are theoretically possible?
How many of these have been tested during evolution? Protein sequence space

Assuming a peptide of length 100 aa

Possible combinations: $n_c = 20^{100} \approx 1.27 * 10^{130}$

Volume of one peptide:

$$r_{atom} \approx 2\text{\AA} v_{atom} \approx 35\text{\AA}^3 packing \approx 75\% v_{peptide} \approx 1.3 * 10^5 \text{\AA}^3$$

Protein sequence space

□ 1.27*10¹³⁰ combinations. For comparison ...

| Volume of the Earth: | $R \approx 6.4 * 10^{3} \text{ km} \approx 6.4 * 10^{16} \text{ Å}$ $V = \frac{4}{3} \pi R^{3} \approx 1.1 * 10^{51} \text{ Å}^{3}$ | |
|----------------------|---|--|
| Peptides/Earth: | $n_p \approx \frac{1.1 * 10^{51}}{1.3 * 10^5} \approx 7.7 * 10^{45}$ | |



□ 1.27*10¹³⁰ combinations. For comparison ...

Age of the Earth: $3 * 10^9$ years $\approx 2.6 * 10^{13}$ hours $\approx 9.5 * 10^{16}$ sec $\approx 9.5 * 10^{28}$ psec

If the whole planet consisted of peptides, and peptides were renewed every psec...

$$n_{\rm T} \approx (7.7 \times 10^{45}) \cdot (9.5 \times 10^{28}) \approx 7.3 \times 10^{74}$$



Assuming a peptide of length 100 aa

Possible combinations: $n_c = 20^{100} \approx 1.27 * 10^{130}$

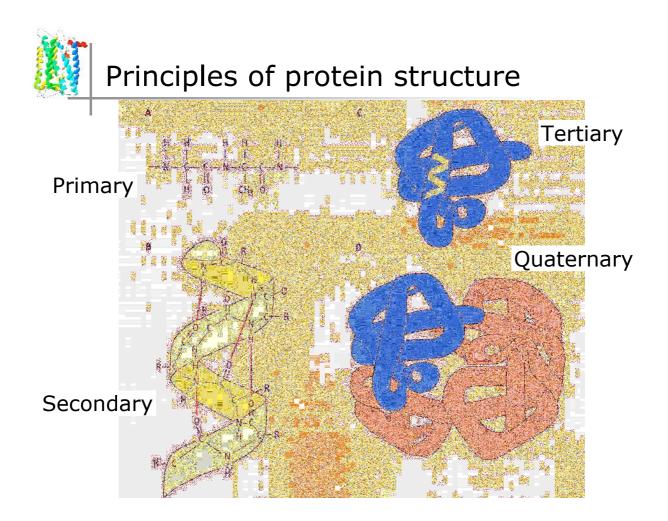
If the whole planet consisted of peptides, and peptides were renewed every psec...

$$n_{\rm T} \approx (7.7 \times 10^{45}) \cdot (9.5 \times 10^{28}) \approx 7.3 \times 10^{74}$$

$10^{130} - 10^{75} \approx 10^{130}$

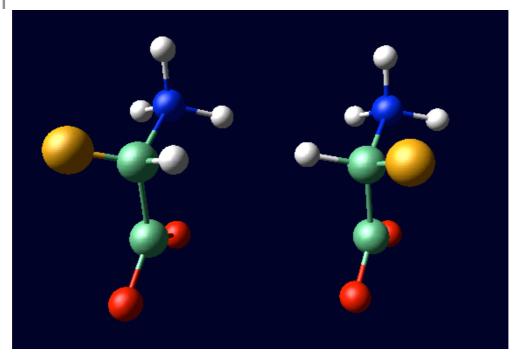


- Primary Structure
- Secondary Structure
- Tertiary Structure
- Quaternary Structure



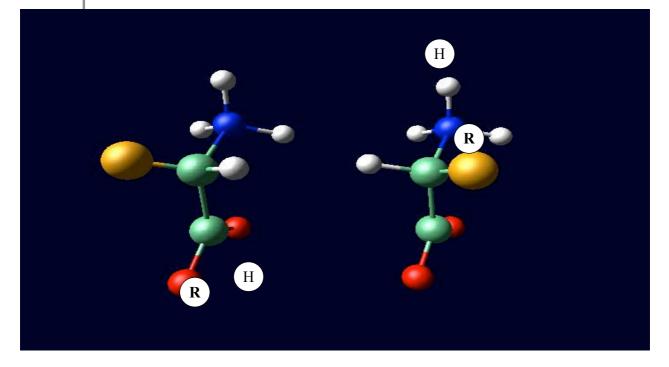


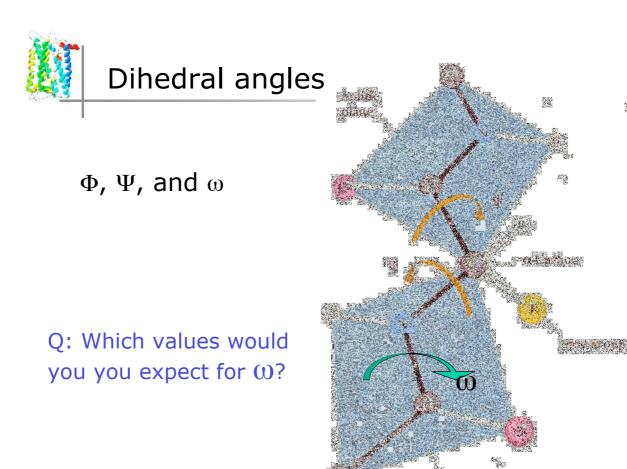
Stereochemistry: L- and D-amino acids





Geometry of a peptide bond

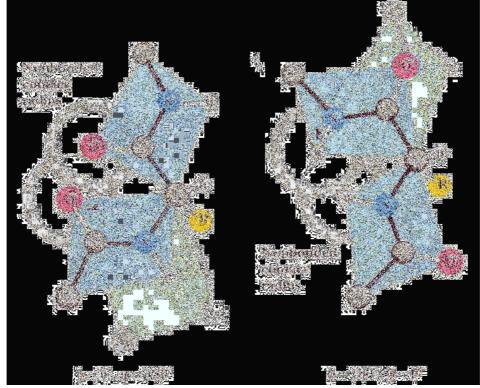




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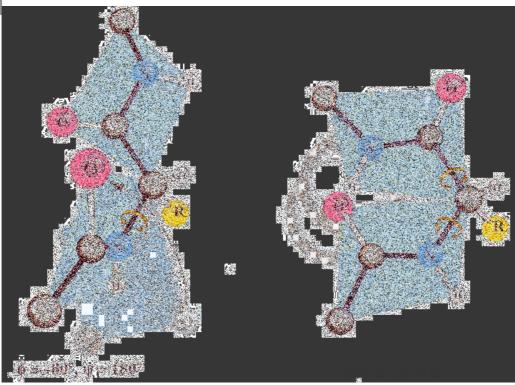
Dihedral angles Φ and Ψ



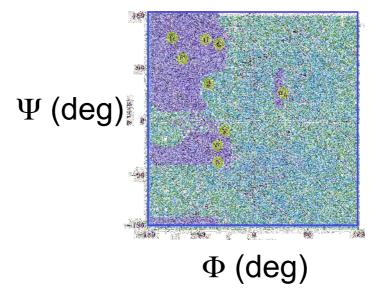
Anastis Press

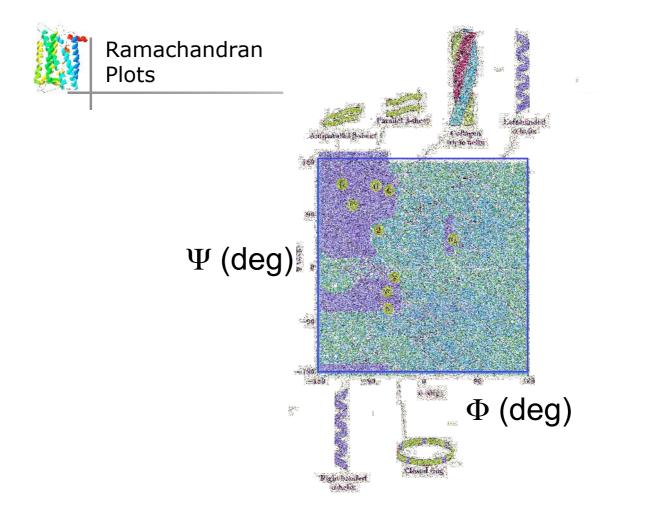


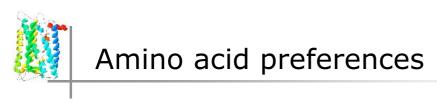
Dihedral angles Φ and Ψ



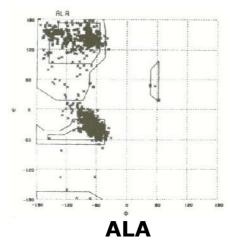


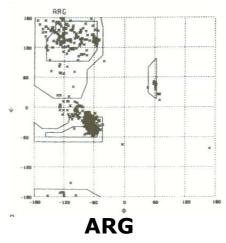






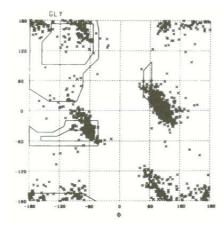
Alanine and Arginine

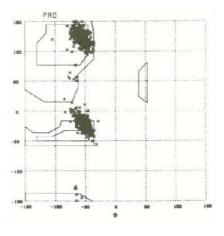






Amino acid with special preferences:

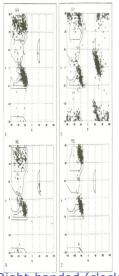




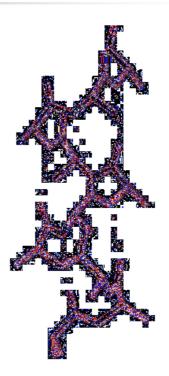
GLY

PRO

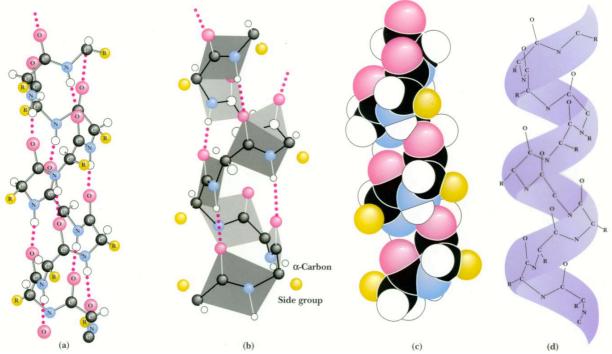




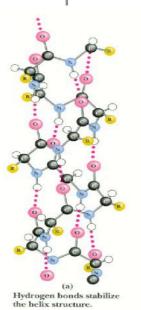
Right-handed (clockwise) helical conformation

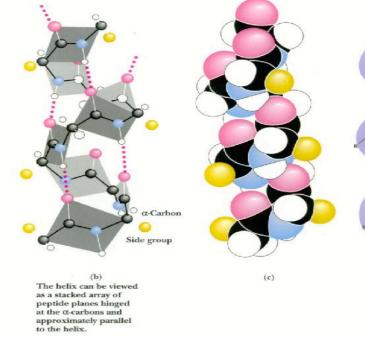








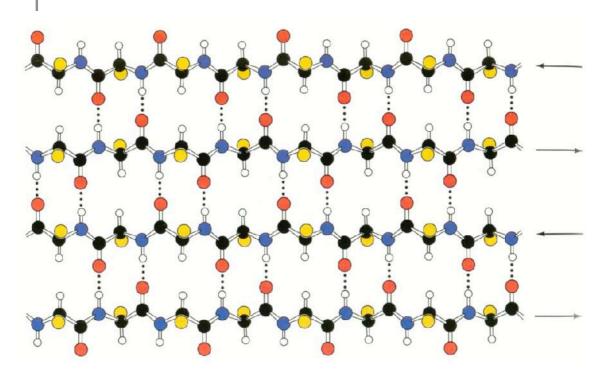


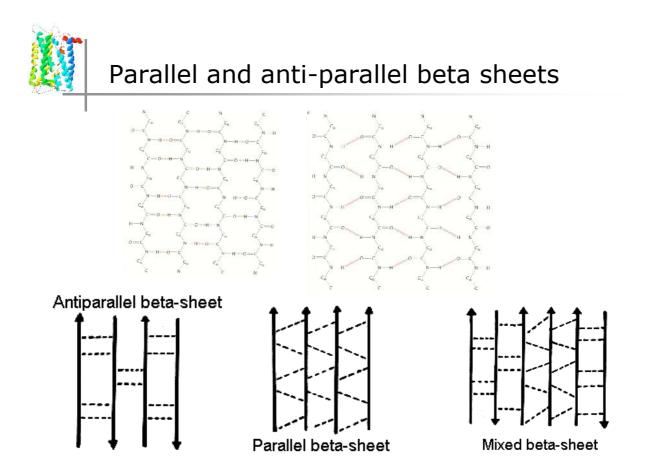


(d)



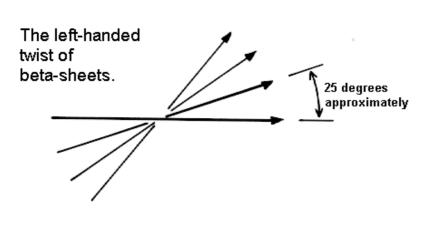
Anti-parallel beta sheet

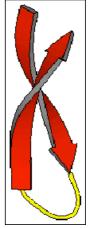






Left-handed twist in beta-sheets

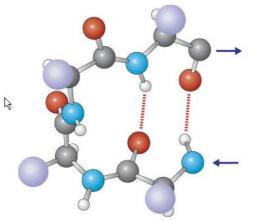




Bovine pancreatic trypsin inhibitor

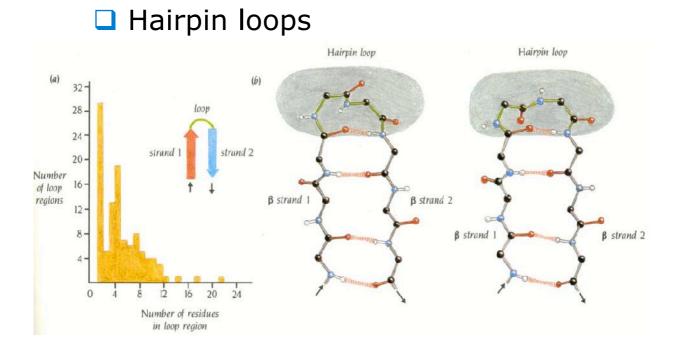
0° - 30° per aa





Schematic diagram showing the interresidue backbone hydrogen bonds that stabilize the reversal of the chain direction. Side chains are depicted as large light purple spheres. Due to the tight geometry of the turn, some residues are found more commonly in turns than others.







Conformational Preferences

| Amino | | Preference | |
|-------|-----------|-----------------|--------------|
| acid | α-helix | β -strand | reverse turn |
| Glu | 1.59 | 0.52 | 1.01 |
| Ala | 1.41 | 0.72 | 0.82 |
| Leu | 1.34 | 1.22 | 0.57 |
| Met | 1.30 | 1.14 | 0.52 |
| GIn | 1.27 | 0.98 | 0.84 |
| Lys | 1.23 | 0.69 | 1.07 |
| Arg | 1.21 0.84 | 0.90 | |
| His | 1.05 | 0.80 | 0.81 |
| Val | 0.90 | 1.87 | 0.41 |
| lle | 1.09 | 1.67 | 0.47 |
| Tyr | 0.74 | 1.45 | 0.76 |
| Cys | 0.66 | 1.40 | 0.54 |
| Trp | 1.02 | 1.35 | 0.65 |
| Phe | 1.16 | 1.33 | 0.59 |
| Thr | 0.76 | 1.17 | 0.90 |
| Gly | 0.43 | 0.58 | 1.77 |
| Asn | 0.76 | 0.48 | 1.34 |
| Pro | 0.34 | 0.31 | 1.32 |
| Ser | 0.57 | 0.96 | 1.22 |
| Asp | 0.99 | 0.39 | 1.24 |

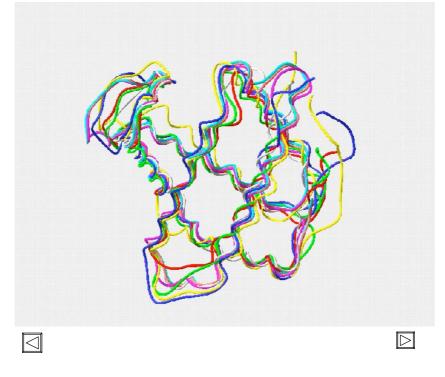
α

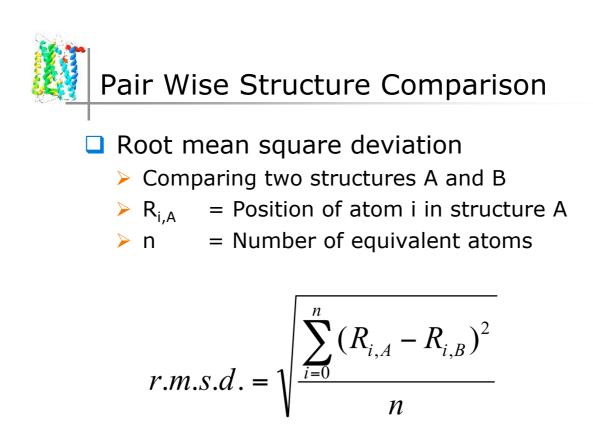
β

RT

Biochimica et Biophysica Acta 916: 200-204 (1987).

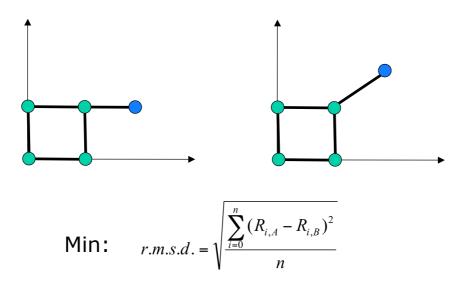


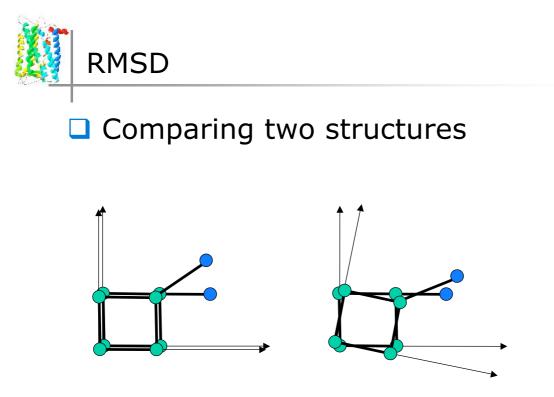






Comparing two structures







•(Iterative) Least squares superposition (e.g. LSQ-Kabsch)

- uses RMSD (root mean square deviation) as similarity measure
- useful if sequence alignment is unambiguous
- only useful for similar structures (rmsd < 5 Å)

•DALI (Distance Matrix Alignment)

- http://www.ebi.ac.uk/dali/
- Represents structures as 2-dimensional array of distances between C α positions
- Compare two structures by overlapping these arrays
- · Similarity expressed as z-score vs. database background distribution
- •CE (Combinatorial Extension of the optimal path)
- http://cl.sdsc.edu/ce.html
- start by finding octameric fragments in both structures that can be overlapped
- extend aligned fragment pairs (AFPs) to find optimal alignment
- · Similarity expressed as z-score vs. database background distribution





Protein Structure is better conserved than sequence

- Structural alignments establish equivalences between amino acid residues based on the 3D structures of two or more proteins
- Structure alignments therefore provide information <u>not</u> available from sequence alignment methods
- Structural alignments can be used to guide sequence alignments (see: T_COFFEE / SAP)



PDB

- EBI-MSD
- SCOP
- CATH







This page contains:

<u>PDB Holdings List</u>: Breakdown of the contents of the PDB, following the latest update <u>PDB Content Growth</u>: Data on the growth in the number of structures in the PDB

PDB Holdings List: 13-Apr-2004

| | | Моlесиlе Туре | | | | |
|-------|-----------------------------------|---------------------------------------|--------------------------------------|---------------|---------------|-------|
| | | Proteins, Peptides, and Viruses | Protein/Nucleic Acid Complexes | Nucleic Acids | Carbohydrates | total |
| Exp. | X-ray Diffraction and other | 19696 | 969 | 727 | 14 | 21406 |
| Tech. | NMR | 3031 | 97 | 577 | 4 | 3709 |
| | Total | 22727 | 1066 | 1304 | 18 | 25115 |



Number of folds is limited

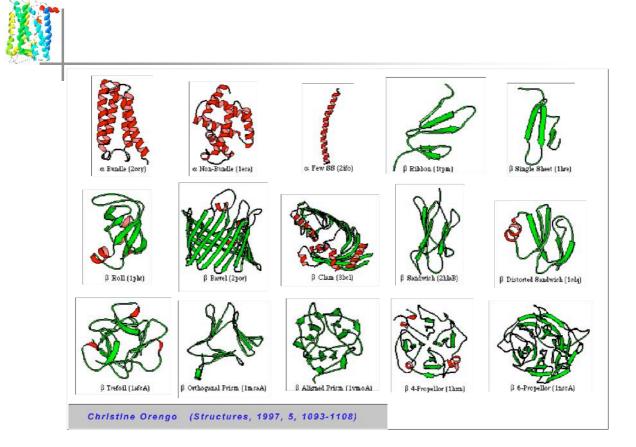
> according to C. Chothia ca. 1000 – 5000 folds

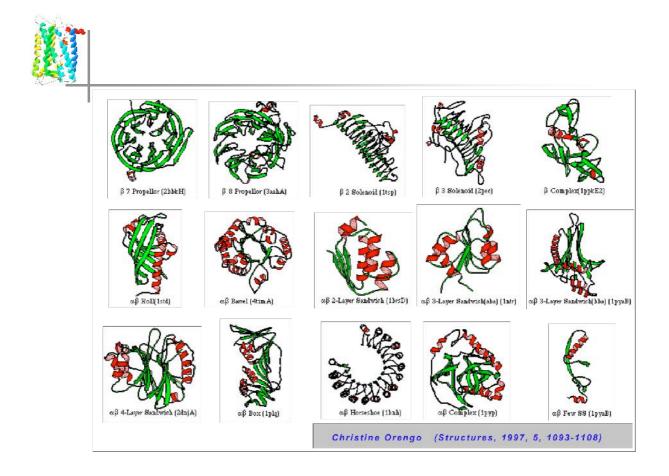
Fold databases systematically classify protein structure folds

FSSP

- > Fold classification based on Structure-Structure alignment of Proteins
- http://www2.ebi.ac.uk/dali/fssp/
- > Contains structurally aligned proteins based in DALI

SCOP

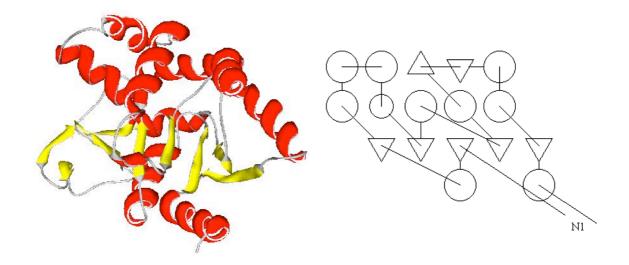






TOPS: Topology of Protein Strcuture

http://www.tops.leeds.ac.uk/





- □ Hierarchical classification of protein domain structures
- UCL, Janet Thornton & Christine Orengo
- clusters proteins at four major levels:
 - Class(C)
 - > Architecture(A)
 - > Topology(T)
 - Homologous superfamily (H)





Class(C)

derived from secondary structure content is assigned automatically

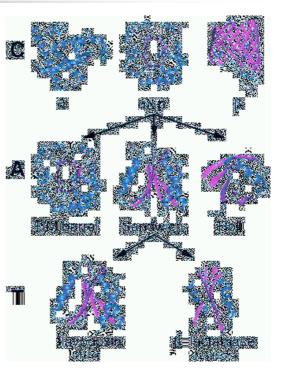
Architecture(A)

describes the gross orientation of secondary structures, independent of connectivity.

Topology(T)

clusters structures according to their topological connections and numbers of secondary structures

[http://www.biochem.ucl.ac.uk/bsm/cath_new/]





- MRC Cambridge (UK), Alexey Murzin, Brenner S. E., Hubbard T., Chothia C.
- □ hierarchical classification of protein domain structures
- created by manual inspection
- comprehensive description of the structural and evolutionary relationships
- organized as a tree structure
 - Class
 - Fold
 - Superfamily
 - > Family
 - Species



- □ PDB: http://www.pdb.org
- □ EBI-MSD http://www.ebi.ac.uk/msd/
- □ SCOP http://scop.mrc-lmb.cam.ac.uk/scop/
- □ CATH http://www.biochem.ucl.ac.uk/bsm/cath_new/



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