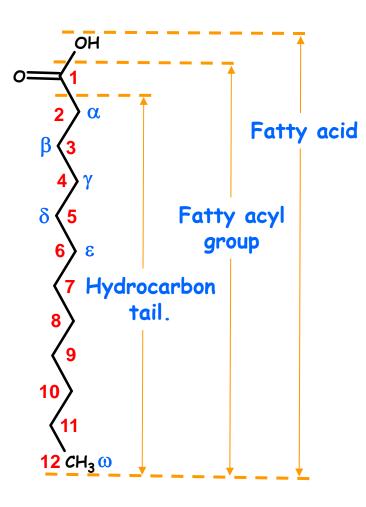
Chapter 11 – Lipids Problems: 2-8,10-12,15-17.

- Lipids are essential components of all living organisms
- Lipids are water insoluble organic compounds
- They are hydrophobic (nonpolar) or amphipathic (containing both nonpolar and polar regions)
  - 1. Free fatty acids
  - 2. Triacylglycerols
  - 3. Phospholipids
  - 4. Glycolipids
  - 5. Steroids

# Fatty Acids



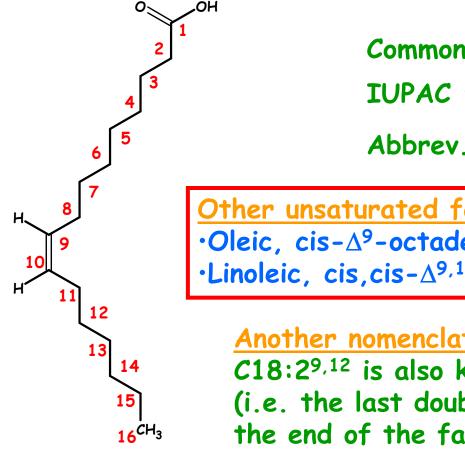
Common = lauric acid

**IUPAC = dodecanoic acid** 

Abbrev. = C12:0

Other fatty acids (examples) •Myristic, tetradecanoic, C14:0 •Palmitic, hexadecanoic, C16:0 •Stearic, octadecanoic, C18:0

# Unsaturated Fatty Acids



Common = palmitoleic acid **IUPAC** =  $cis - \Delta^9$ -hexadecenoic acid

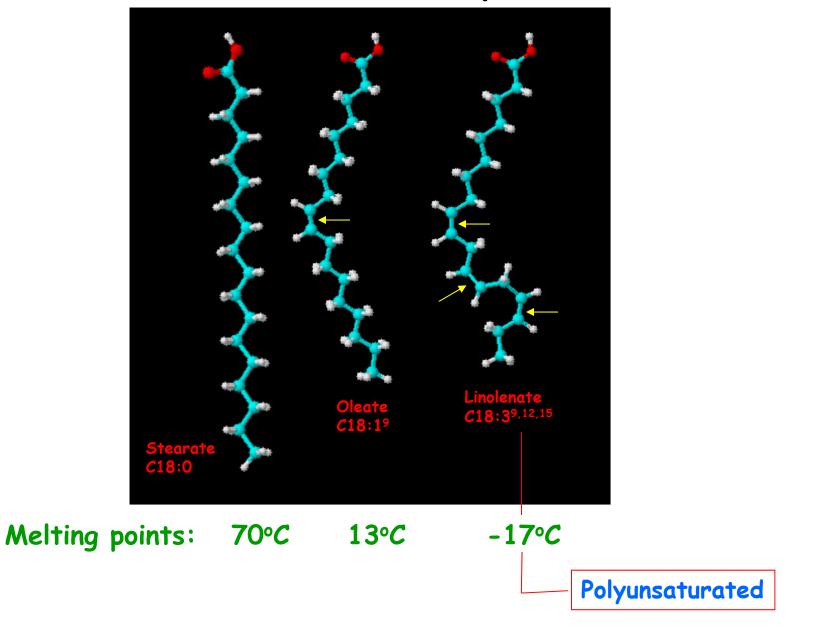
Abbrev. =  $C16:1^9$ 

Other unsaturated fatty acids (examples) •Oleic, cis- $\Delta^9$ -octadecenoic, C18:1<sup>9</sup> •Linoleic, cis, cis- $\Delta^{9,12}$ -octadecadienoic, C18:2<sup>9,12</sup>

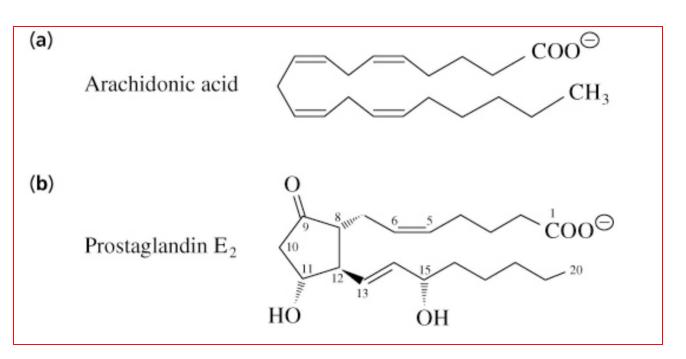
#### Another nomenclature system:

C18:2<sup>9,12</sup> is also known as an  $\omega$ -6 fatty acid (i.e. the last double bond is 6 carbons from the end of the fatty acid chain.)

# Unsaturated Fatty Acids

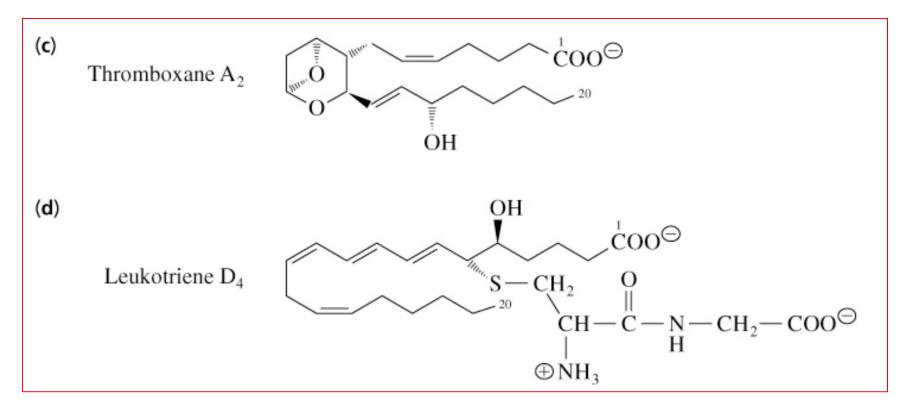


# Eicosanoids



- Eicosanoids are oxygenated derivatives of  $C_{20}$  polyunsaturated fatty acids (e.g. arachidonic acid)
- Prostaglandins eicosanoids having a cyclopentane ring
- <u>Aspirin</u> alleviates pain, fever, and inflammation by inhibiting the synthesis of prostaglandins

### Other Eicosanoids



- $\cdot$  Prostaglandin E<sub>2</sub> can cause constriction of blood vessels
- Thromboxane  $A_2$  involved in blood clot formation
- Leukotriene  $D_4$  mediator of smooth-muscle contraction and bronchial constriction seen in asthmatics

Number of carbon atoms	Number of double bonds	Common name	Systematic name	Formula
→ 12	0	Laurate	n-Dodecanoate	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>10</sub> COO⁻
→ 14	0	Myristate	<i>n</i> -Tetradecanoate	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>12</sub> COO <sup>-</sup>
	0	Palmitate	<i>n</i> -Hexadecanoate	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>14</sub> COO⁻
→ 18	0	Stearate	n-Octadecanoate	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>16</sub> COO⁻
20	0	Arachidate	<i>n</i> -Eicosanoate	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>18</sub> COO⁻
22	0	Behenate	<i>n</i> -Docosanoate	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>20</sub> COO <sup>-</sup>
24	0	Lignocerate	<i>n</i> -Tetracosanoate	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>22</sub> COO <sup>-</sup>
→ 16	1	Palmitoleate	<i>cis</i> - $\Delta^9$ -Hexadecenoate	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>5</sub> CH==CH(CH <sub>2</sub> ) <sub>7</sub> COO <sup>-</sup>
→ <sup>18</sup>	1	Oleate	<i>cis</i> - $\Delta^{9}$ -Octadecenoate	CH,(CH,),CH==CH(CH,),COO-
18	2	Linoleate	<i>cis, cis</i> - $\Delta^9$ , $\Delta^{12}$ - Octadecadienoate	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>4</sub> (CH=CHCH <sub>2</sub> ) <sub>2</sub> (CH) <sub>6</sub> COO <sup>-</sup>
18	3	Linolenate	all- <i>cis</i> -Δ°, Δ <sup>12</sup> , Δ <sup>15</sup> - Octadecatrienoate	CH <sub>3</sub> CH <sub>2</sub> (CH <sup>−−</sup> CHCH <sub>2</sub> ) <sub>3</sub> (CH <sub>2</sub> ) <sub>6</sub> COO <sup>−</sup>
20	4	Arachidonate	all- <i>cis</i> ∆⁵, ∆ <sup>8</sup> , ∆¹¹, ∆¹⁴- Eicosatetraenoate	$CH_3(CH_2)_4(CH = CHCH_2)_4(CH_2)_2COO^-$

#### Table 11.1 Some naturally occurring fatty acids in animals

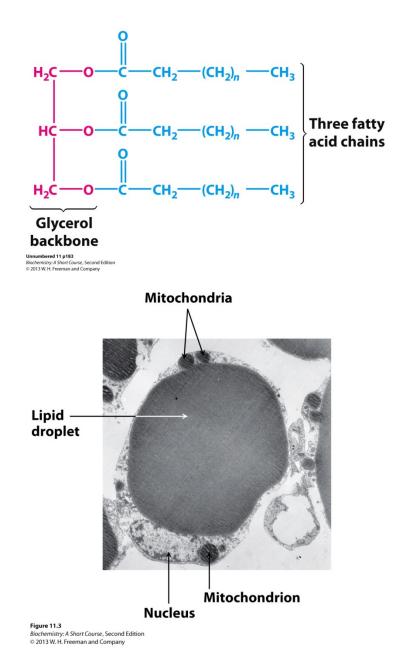
Table 11.1

*Biochemistry: A Short Course*, Second Edition © 2013 W. H. Freeman and Company

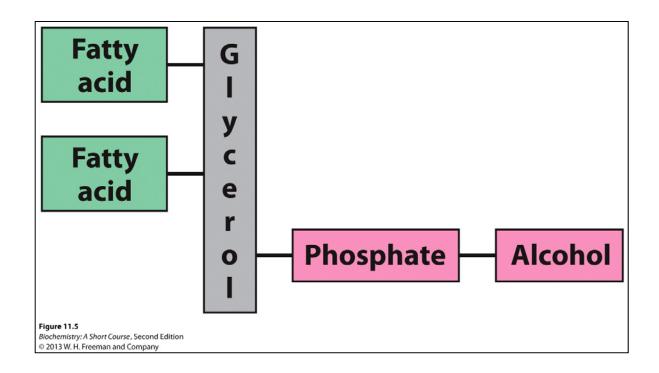
# Triacylglycerol

- Fatty acids are important metabolic fuels (2-3 times the energy of proteins or carbohydrates)
- Fatty acids are stored as neutral lipids called triaclyglycerols (TGs)
- TGs are composed of 3 fatty acyl residues esterified to a <u>glycerol</u> (3carbon sugar alcohol)
- TGs are very hydrophobic, and are stored in cells in an anhydrous form (e.g. in fat droplets)

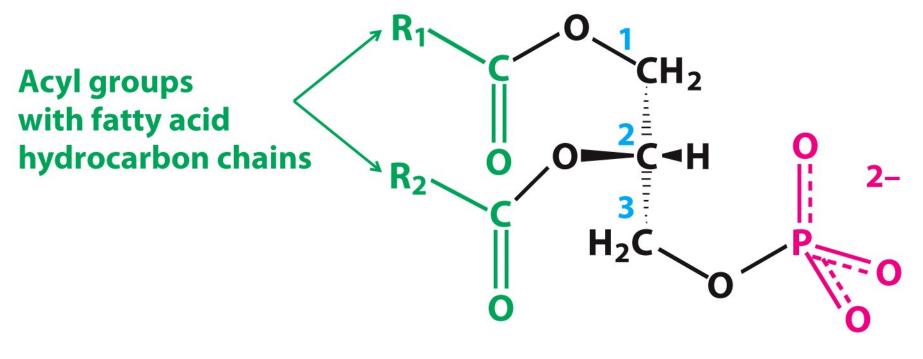
TGs are catabolized by lipases. Digestion requires bile salts (solubilize TGs). Transport of TGs is accomplished through lipoproteins.



# Phospholipids



- The most abundant lipids in membranes
- Possess a glycerol backbone
- A phosphate is esterified to both glycerol and another compound bearing an -OH group
- <u>Phosphatidates</u> are glycerophospholipids with two fatty acid groups esterified to C-1 and C-2 of glycerol 3-phosphate



#### Phosphatidate (Diacylglycerol 3-phosphate)

Figure 11.6 Biochemistry: A Short Course, Second Edition © 2013 W. H. Freeman and Company

### Types of phospholipids

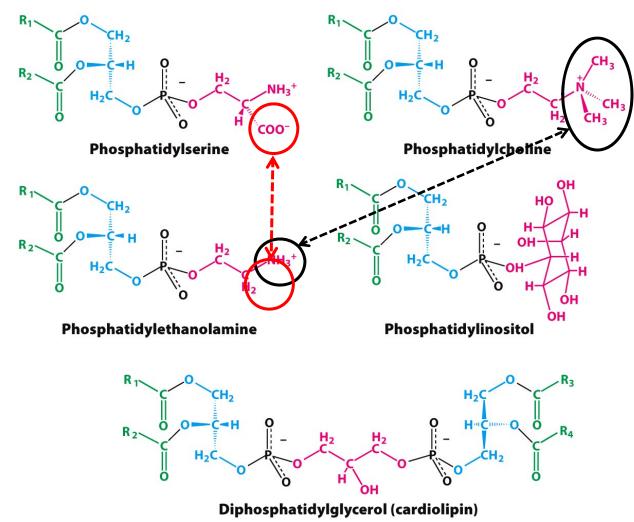
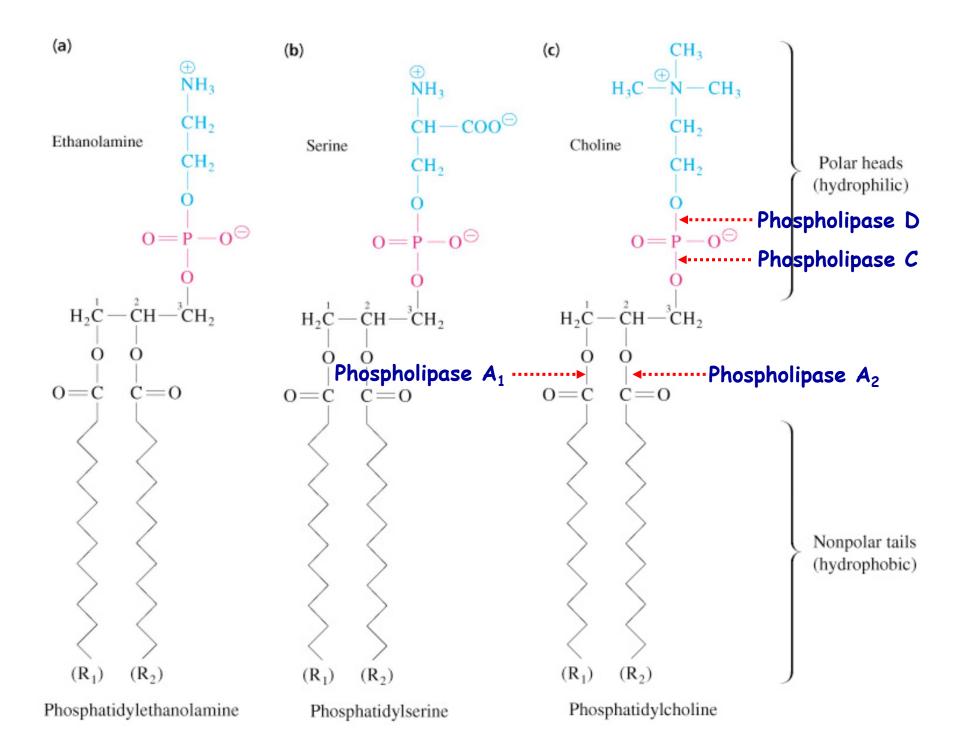
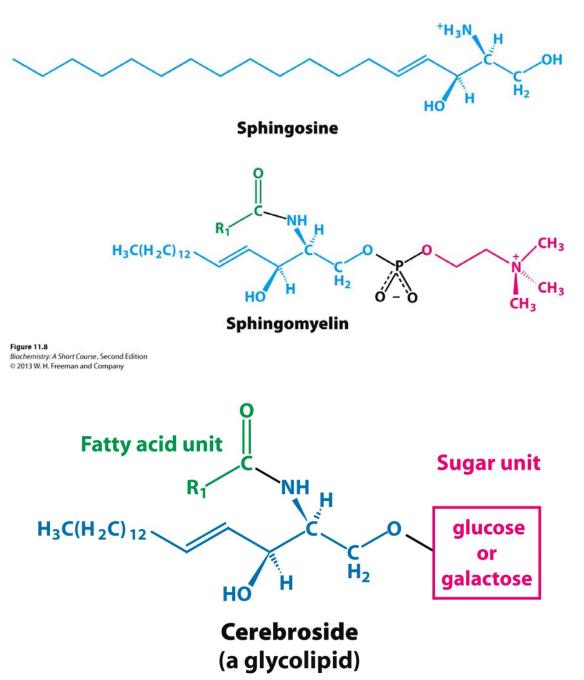


Figure 11.7 Biochemistry: A Short Course, Second Edition © 2013 W. H. Freeman and Company



#### Sphingolipids



Unnumbered 11 p186a Biochemistry: A Short Course, Second Edition © 2013 W. H. Freeman and Company

# Steroids

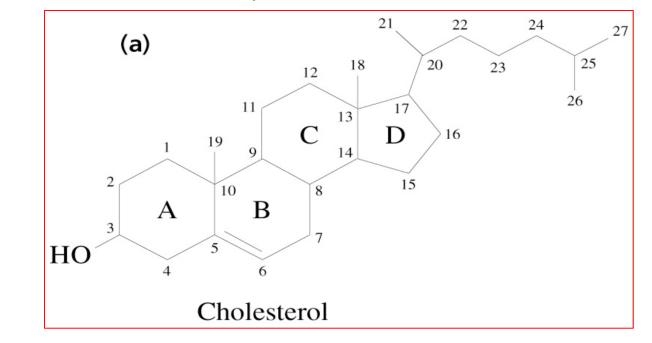
- Classified as isoprenoids related to 5-carbon isoprene (found in membranes of eukaryotes)
- Steroids contain four fused ring systems: 3-six carbon rings (A,B,C) and a 5-carbon D ring  $H_3C$
- Ring system is nearly planar

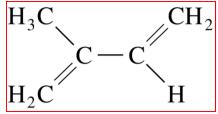
-OH = "ol"

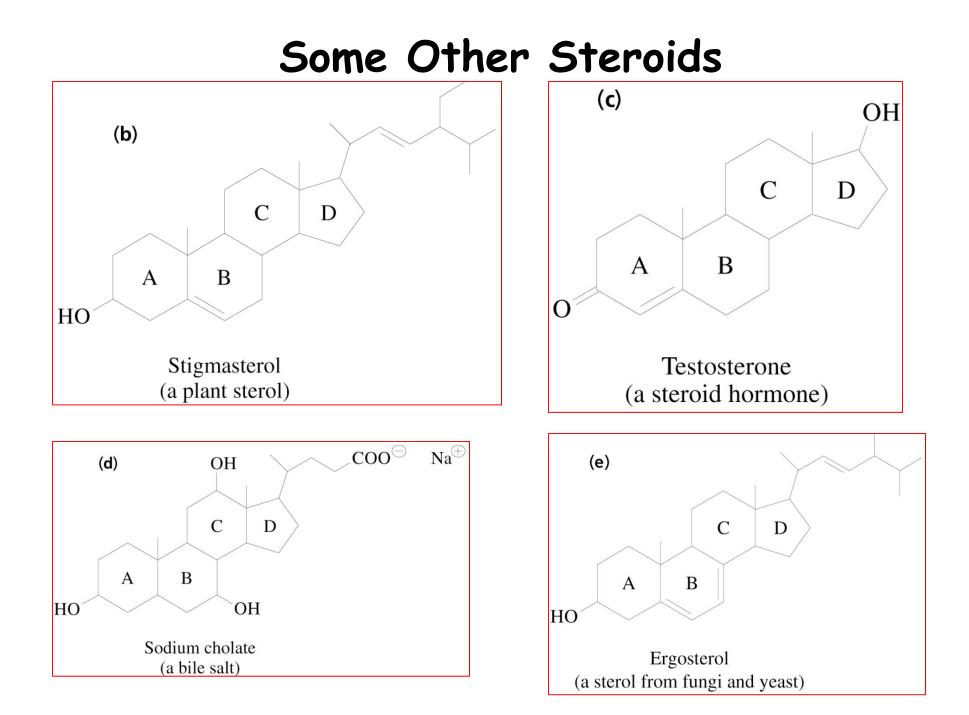
-C=O = "one"

-COOH = "ate"

• Substituents point either down (a) or up (b)







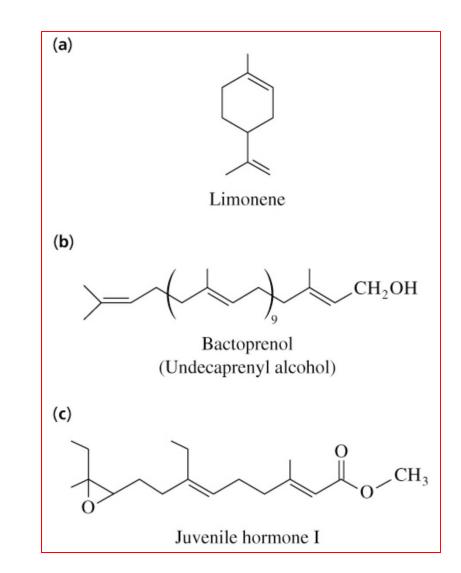
### Functions and Properties of Cholesterol

- Cholesterol modulates the fluidity of mammalian cell membranes
- It is also a precursor of the steroid hormones and bile salts
- It is a sterol (has hydroxyl group at C-3)
- The fused ring system makes cholesterol less flexible than most other lipids
- Cholesterol is converted to cholesteryl esters for cell <u>storage</u> or <u>transport</u> in blood
- Fatty acid is esterified to C-3 OH of cholesterol
- Cholesterol esters are very water insoluble and must be complexed with phospholipids or amphipathic proteins for transport

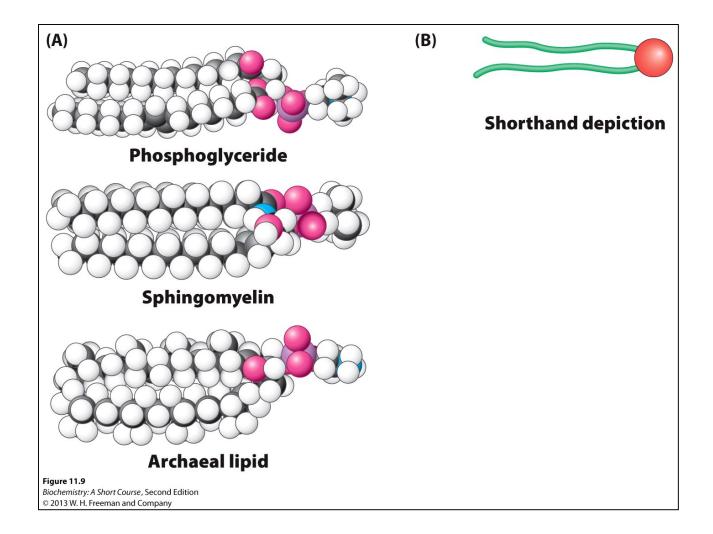
### Some Other Isoprenoids

Lipid vitamins

 (A, D, E, and K)
 are isoprenoid
 derivatives



# Membrane Lipids are Amphipathic



## Some Membrane Proteins are Lipid Anchored

