INHALANTS

Oxygen

 O_2

MOP:

by heating Potassium chlorate

• 2KClO₃ _____ 3O₂ + KCl

Physical Properties:

- Colorless gas
- Odourless
- Tasteless
- Soluble in water
- Soluble in alcohol
- Freely soluble in silver solution
- Slightly heavier than air

Chemical Properties:

1. It react with carbon form carbon dioxide

$$C + O_2 \longrightarrow CO_2$$

2. It react with sulpher form sulpher dioxide

$$S + O_2 \longrightarrow SO_2$$

Uses:

- 1. Prevention & Treatment of hypoxia.
- 2. Dilatants for anesthetic gases.
- 3. Treatment of carbon monoxide poisoning.
- 4. Treatment of severe pulmonary damage.
- 5. Liquid oxygen used as fuel in rocket technology.



Storage

"Oxygen is stored and supplied in metal cylinders. The shoulder of cylinder are painted with white color & remaining with black color."

Oxygen cylinder

