Determination of Radiation Constant of Iron Cylinder

Aim: To determine the radiation constant of iron cylinder

Requirements: Iron cylinder, thermometer, wooden plank, stop clock, tongs & heating source

Principle: Heat is lost from hot cylinder surface to surrounding atmosphere by means of conduction and radiation. Heat transfer by radiation occurs, energy transfer through space by means of electromagnetic radiation (waves). Thus a body acts as an emitter, than energy being transmitted through the intervening spaces, and it is effective even in a perfect vacuum or interspacious space. The amount of thermal energy radiated by a surface is increased rapidly with increasing temperature, when a heat flows by actual mixing of warmer portions with cooler portions of the same material. This mechanism is known as **Convection**

Procedure:

1. Select an iron cylinder whose surface is smooth & radiation constant is determined.

2. Note & record the room temperature, measure the weight, area and radius of the cylinder. Insert the thermometer into the cavity of the cylinder.

3. Plcace the cylinder on a tripod stand and heats it for about 300° C

4. Now hold the cylinder with the tongs and place on a non-conducting surface (wooden), without touching any surface.

5. Note the temperature reading for every 5 min, by using stop clock.

6. Plot a graph between temperature on y-axis and time on x-axis.

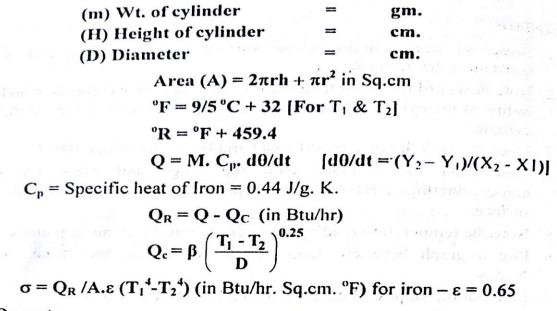
7. Find out the slopes $d\emptyset/dt$, at various arbitrary temperatures.

8. Caliculate the radiation constant by using equation.

Precautions:

1. The cylinder must be transferred as fast as possible

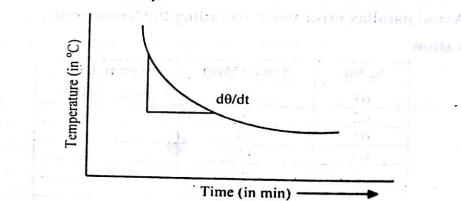
2. Avoid parallax error while recording the temperature.



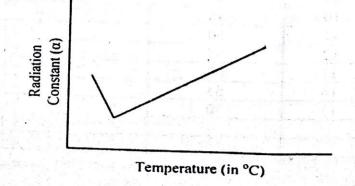
Report :

Graphs : Graphical representation

(a) Determination of slope



(b) Influence of temperature on radiation constant



The rate of heat transfer by a hot body equal to sum of heat transfer through conduction and radiation

$$Q = Q_{\rm C} + Q_{\rm R} \qquad \dots \dots (1)$$

Where, $\mathbf{Q} = \text{Rate of heat loss/transfer in Btu/hr}$

 Q_C = Rate of heat transfer by conduction in **Btu/hr**

 Q_R = Rate of heat transfer by radiation in **Btu/hr**

The rate of heat transfer through a heat body to 0 °C per time

$$\mathbf{Q} = \mathbf{M}\mathbf{C}_{\mathbf{P}} \, \mathbf{d}\boldsymbol{\theta}/\mathbf{d}\mathbf{t} \qquad \dots (2)$$

Where, M = Mass of the cylinder in gm.

 $C_p =$ Specific heat of material (Cylinder) in Cal/gm/°C

 $d\theta/dt =$ Slope of the graph plotted between time on X-axis and temperature on Y-Axis

Q = Rate of heat transfer in Btu/hr

The rate o _____ ransfer through radiation is

$$Q_{R} = \epsilon A \sigma (T_{1}^{4} - T_{2}^{4})$$
(3)

(Stefan-Boltzmann's Law gives the total amount of radiation emitted by a body)

Where,

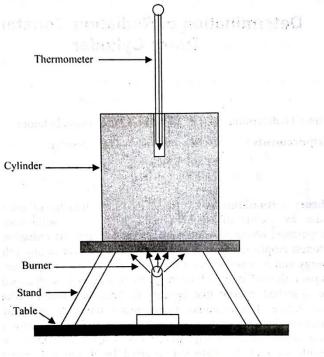
- $\varepsilon = \text{Emissivity of the cylinder (for black body } \varepsilon = 1)$
- A= Area of the cylinder $[2\pi rh + \pi r^2]$ in Sq.cm.
- σ = Stepnan- Boltzman constant (for black bodies b = 0.174 × 10⁻⁸ Btu/hr. Sq. ft. °F⁴ or 5.67 × 10⁻⁸ W/m².K⁴)
- 1 & T₂ are the absolute temperatures of initial & final in °C respectively.

Heat loss through conducts :

$$Q_{c} = \beta \left[\frac{T_{1} - T_{2}}{D} \right]^{0.25} \dots (4)$$

 β = Convection Constant (Iron = 0.28)

 $\mathbf{D} = \text{Diameter of cylinder}$



Experimental setup for Determination of Radiation constant of Iron cylinder.

Observation:

S.No	Time(min)	Temp.(°c)
1	0	Temp.(°c)Above 300°C
2	5	
15	70	

Bibliography:

- 1. Walter L.Badger and Juliust T.Banchero(page no.117 & 152-155)
- 2. K.Sambamurthy (page no 75-77)

Determination of Radiation Constant of Brass Cylinder

Aim: To determine the radiation constant of Brass cylinder

Requirements: Brass cylinder, thermometer, wooden plank, stop clock, tongs & heating source

Principle: Heat is lost from hot cylinder surface to surrounding atmosphere by means of conduction and radiation. Heat transfer by radiation occurs, energy transfer through space by means of electromagnetic radiation (waves). Thus a body acts as an emitter, than energy being transmitted through the intervening spaces, and it is effective even in a perfect vacuum or interspacious space. The amount of thermal energy radiated by a surface is increased rapidly with increasing temperature, when a heat flows by actual mixing of warmer portions with cooler portions of the same material. This mechanism is known as **Convection**

Procedure:

1. Select an brass cylinder whose surface is smooth & radiation constant is determined.

2. Note & record the room temperature, measure the weight, area and radius of the cylinder. Insert the thermometer into the cavity of the cylinder.

3. Place the cylinder on a tripod stand and heats it for about 300° C

4. Now hold the cylinder with the tongs and place on a non-conducting surface (wooden), without touching any surface.

5. Note the temperature reading for every 5 min, by using stop clock.

6. Plot a graph between temperature on y-axis and time on x-axis.

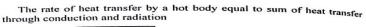
7. Find out the slopes $d\emptyset/dt$, at various arbitrary temperatures.

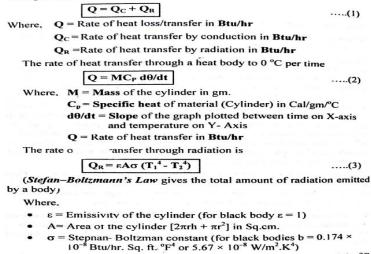
8. Caliculate the radiation constant by using equation.

Precautions:

1. The cylinder must be transferred as fast as possible

2. Avoid parallax error while recording the temperature.





1, & T₂ are the absolute temperatures of initial & final in °C respectively.

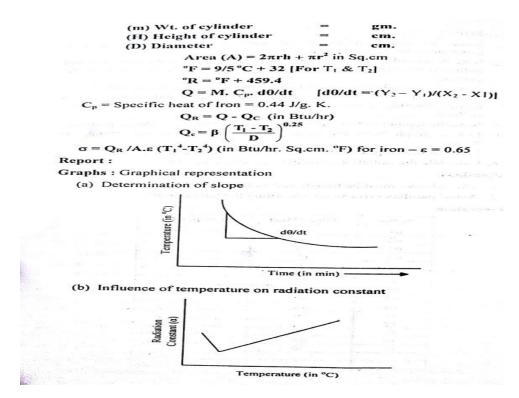
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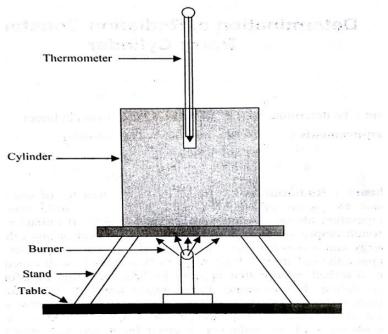
Heat loss through conducts :

$$\mathbf{Q}_{c} = \beta \left[\frac{\mathbf{T}_{1} - \mathbf{T}_{2}}{\mathbf{D}} \right]^{0.25}$$

 β = Convection Constant (Iron = 0.28)

 $\mathbf{D} = \mathbf{D}$ iameter of cylinder





Experimental setup for Determination of Radiation constant of Iron cylinder.

Observation:

S.No	Time(min)	Temp.(°c)Above 300°C
1	0	Above 300 ^o C
2	5	
15	70	

Bibliography:

- 1. Walter L.Badger and Juliust T.Banchero(page no.117 & 152-155)
- 2. K.Sambamurthy (page no 75-77)