THIAZOLE

Thiazole is an aromatic five membered heterocyclic compound having sulphur and ni-Thiazole is a position. Its molecular formulae is C₃H₃SN. It is present in various natural trogen at 1 and 3 position. The structure of thiazole is: products. The structure of thiazole is:

Methods of Preparation Hantzsch Synthesis: It is also known as synthesis from α-halocarbonyl compounds. This method involves reaction of α -haloketones with thioamides to give thiazole. 1.

R-C-CHR + NH₂-C-R
$$\xrightarrow{R}$$
 \xrightarrow{R} \xrightarrow{R} \xrightarrow{N} \xrightarrow{R} \xrightarrow

For example:

$$PhCOOCH_2Br + PhCSNH_2$$
 Ph

Reaction with Cyanoamidine: This reaction involves C-C bond formation and also 2. known as 1, 3 Dipolar addition.

$$NH_2-C-S-CH_2-COPh$$
 $NAOEt$
 NH_2
 NCN
 $NAOEt$
 NH_2
 NH_2

By rearrangement of α-thiocyanoketones : Cyclisation of α-thiocyano ketones in aqueous concentrated sulphuric acid forms substituted thiazole.

$$R_{1} - C - C - S - C \equiv N \xrightarrow{H_{3}O^{+}} R_{1} \xrightarrow{R_{1}} OH$$

$$\downarrow \qquad \qquad \downarrow \qquad$$

Cook-Heilbron's Synthesis: In this method Carbon disulphide reacts with α-4. aminonitriles to form substituted thiazoles.

$$R - C - C \equiv N + CS_2$$

$$NH_2 \qquad Carbon disulphide \qquad NH_2 \qquad SH$$

$$Thiazole$$

Gabriel's synthesis: It involves reaction of acylaminoketone with an equimolar amount of PS₅ to give 2-phenyl-5-alkyl thiazole. 5.

$$R_{3} - C - C - NH - C - R_{1}$$

$$R_{3} - C - C - NH - C - R_{1}$$

$$R_{3} - C - C - NH - C - R_{1}$$

$$R_{3} - C - C - NH - C - R_{1}$$

$$R_{3} - C - C - NH - C - R_{1}$$

$$R_{3} - C - C - NH - C - R_{1}$$

$$R_{3} - C - C - NH - C - R_{1}$$

From Vinyl Bromide: Substituted thiazoles can be prepared by intramolecular nucleophillic substitution of vinyl bromides. 6.

Chemical Reactions of Thiazole

Nucleophillic attack at C-2: Nucleophillic attack on the ring carbon atoms occurs readily with thiazole. Sodamide converts thiazole into 2-aminothiazole. 1.

$$\begin{array}{c}
N_{\text{aNH}_2} \\
N_{\text{aNH}_2}
\end{array}$$
Thiazole

2-Aminothiazole

Thiazole having a leaving group at the C-2 position reacts readily with nucleophiles to give displacement products.

$$\begin{bmatrix}
N \\
S
\end{bmatrix}$$

$$C1$$

$$NH_{2}$$

$$NH_{2}$$

$$C_{2}H_{5}\bar{O}Na^{+}$$

$$NO_{2}$$

$$C_{2}H_{5}\bar{O}Na^{+}$$

$$C_{2}H_{5}\bar{O}Na^{+}$$

Electrophillic substitution reaction: Electrophillic substitution occurs at C-2 and C-5 but substitution at C-2 would be unfavourable relative to attack at C-5, because charge formed in the intermediate cation has to be stabilized by delocalisation through the imine nitrogen.

Thiazole

$$SO_3H$$
 SO_3H
 SO_3H

$$\begin{array}{c|c}
 & N \\
 & N \\$$

3. Alkylation at Nitrogen: Alkylation of thiazoles at nitrogen forms a thiazolium cations which are used as a catalyst in various reactions.

MEDICINAL USES OF THIAZOLE

Thiazole derivatives has a wide variety of applications ranging from bacteriostatics, antibiotics, CNS regulators, High Ceiling Diuretics etc. Thiazoles are also useful for the treatment of inflammation, Hypertension & HIV infections. Some Aminothiazoles are ligands of estrogen receptors as well as adenosine receptor antagonists. Some thiazole analogues are used as fungicides, herbicides or as schistosomicidal and anthelmintic drugs.

Some of comon drugs having thiazole nucleus and their uses are as follows:

- Thiamine also known as vitamin B₁
- 2. All the Penicillins antibiotics contains thiazole ring
- Sulphathiazole (Antibiotic)
- Fenetizole (Antinflammatory)
- Combendazole (Fungicidal)
- 6. Niridazole (Schistozomicidal)
- 7. Aminothiazole is used as a thyroid inhibitor in the treatment of hyperthyroidism
- 8. Amiphenazole is used as a respiratory stimulant
- 9. Brecanavir is useful as anti HIV
- 10. Cefixime and other Cephalosporin antibiotics also belongs to this class.
- 11. Alagebrium is effectie in reducing systolic blood pressure.
- 12. Abafungin is a broad spectrum antifungal agent.