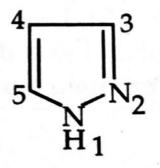
PYRAZOLE

It is an organic compound having molecular formulae C₃H₃N₂H. It is a heterocyclic compound having two nitrogen atoms at position 1 and 2. It is weakly basic in nature.



Methods of Preparation:

By condensing malondialdehyde diethyl acetal with hydrazine.

$$\begin{array}{c|c} CH(OEt)_2 \\ CH_2 \\ CH_2 \\ CH(OEt)_2 \\ \end{array}$$

$$\begin{array}{c|c} NH_2 \\ NH_2 \\ \end{array}$$

$$\begin{array}{c|c} NH_2 \\ Pyrazole \\ \end{array}$$

$$\begin{array}{c|c} Pyrazole \\ \end{array}$$

Tetraethoxypropane

By passing acetylene gas into a cold solution of diazomethane.

By reaction of α,β unsaturated ketones or aldehydes with hydrazine.

Knorr Pyrazole Synthesis: It is an organic reaction used to convert a hydrize or its derivative and a 1,3-dicarbonyl compound into a pyrazole using an acid catalyst. 4.

The mechanism of reaction involves an acid catalyzed imine formation while in hydrazine derivative the attack can happen on either carbonyl carbon and results in 2 possible products. The other nitrogen of hydrazine derivative then attacks the other carbonyl group which has also been protonated by acid and forms a second imine group. This dimine compound gets deprotonated to regenerate the acid catalyst to form the final pyrazole product. For example:

$$CH_3 \longrightarrow C=O$$

$$CH_2 \longrightarrow C_6H_5$$

$$C=O$$

$$CH_2 \longrightarrow NH-NH_2$$

$$C=O$$

$$CH_3 \longrightarrow C_6H_5$$

$$C_6H_5 \longrightarrow C_6H_5$$

Benzoyl acetone

Chemical reations of Pyrazole:

1. Reaction at nitrogen: Hydrogen atom in the -NH group can be replaced easily to form alkylated derivatives. Pyrazole is weakly basic & can form salts with inorganic acids. The imino hydrogen can be replaced by an acyl group.

2. Electrophillic substitution reaction: Electrophillic substitution occurs at 4th position if it is vacant.

$$\begin{array}{c} & & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & & \\ & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\$$

3. Action of Bases: Strong bases on reaction with pyrazole gives Pyrazolyl anion.

$$\begin{array}{c|c}
 & CH_3ONa \\
\hline
 & N \\
\hline
 & MgCl$$

4. Oxidation and Reduction: Pyrazole is resistant to oxidising and reducing agents but gets hydrogenated catalytically, first to pyrazoline and then into pyrazolidine,

5. Iodination: Pyrazole undergoes base catalyzed iodination at position-4.

$$\begin{array}{c|c}
 & I_2 \\
 & N_{AOH}
\end{array}$$

$$\begin{array}{c}
 & I_2 \\
 & N_{AOH}
\end{array}$$

MEDICINAL USES OF PYRAZOLE

Pyrazole and its derivatives have pharmacological effects on humans. Pyrazole moiety have various medicinal values. Some of them are :

- As antimicrobials: Some pyrazole derivatives shows good activity against E.Coli,
 S.Aureus and S.Lactus. These also show good antibacterial activity for e.g. Cefoselix.
- 2. Analgesic agents: Series of Pyrazole derivatives shows anti-inflammatory and analgesic activity. These drugs acts by inhibiting COX-2-enzyme. For example: Phenylbutazone, Feprazone.
- Anticancer agents: Pyrazole and pyrazoline fused systems shows anticancer activity. These have potent cytotoxicity against some tumor cells e.g. Dacarbazine, Azathioprime.
- 4. Antihelmintic activity: For example Albendazole is active against earthworms and Perituma poshuma. For e.g. Niridazole Thiabendazole
- 5. Antioxidant : Some of the pyrazole derivatives have antioxidant activity also.
- ACE Pyrazole derivatives inhibits Angiotensin Converting Enzyme (ACE) and hence helps in lowering blood pressure.
- Antique against Hepatitis A virus and Herpes Simplex virus type-I.